

# Circular Distribution of Corona Current of Multiple-Conductor Transmission Line (Ⅴ)

Saburo MUTO and Masaharu UDAKA

Department of Electrical Engineering

(Received September 11, 1969)

The author investigates the corona starting voltages and the circular distribution of the corona current around a multiple-conductor transmission line in compressed  $\text{SF}_6$ ,  $\text{c-C}_4\text{F}_8$ ,  $\text{C}_2\text{F}_6$  and  $\text{N}_2\text{O}$  gases.

The various characteristics of the d.c. corona current in fluoride gases mixed with  $\text{N}_2\text{O}$  and the results of the gaschromatographic analysis of the decomposition products by corona discharge are described.

## 1. Introduction

The authors have investigated the corona current distribution around the double-conductor and already published the five papers about the investigations.

The 1st, 2nd and 3rd papers<sup>1)~3)</sup> dealt with the various characteristics of the a.c. and d.c. corona current around the multiple-conductor transmission line in air. In the 4th and 5th papers,<sup>4)5)</sup> we described the corona starting voltages, the d.c. corona current distribution and the directivity characteristics around the double-conductor in  $\text{N}_2$ ,  $\text{CO}_2$ ,  $\text{SF}_6$  and the organic fluoride gases like  $\text{c-C}_4\text{F}_8$  and  $\text{C}_2\text{F}_6$ .

In this paper, the investigations about the various characteristics of the d.c. corona discharge in the fluoride gases like  $\text{c-C}_4\text{F}_8$ ,  $\text{C}_2\text{F}_6$  and  $\text{SF}_6$  mixing with  $\text{N}_2\text{O}$  are carried out. And the results of the gaschromatographic analysis of the decomposed products are shown.

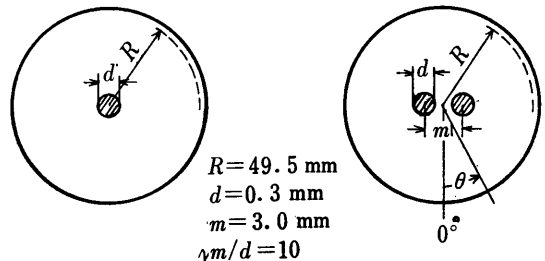
When the perfluorocarbons are exposed to the corona discharge or the arc discharge, they can easily be decomposed.<sup>6)</sup> And, on that account, carbon decomposed is accumulated on the electrodes. It is also reported that  $\text{N}_2\text{O}$  is available to prevent the carbon accumulation.

The authors investigate how the corona characteristics in perfluorocarbon are influenced by mixing with  $\text{N}_2\text{O}$  and, from the viewpoint of the electron attachment coefficient, the corona starting voltages in  $\text{SF}_6$  mixing with  $\text{N}_2\text{O}$  are studied.

## 2. Experimental Equipments, Measuring Method and Definition of Directivity Factor $\kappa$

The experimental equipments in this paper are the same as those in the previous,<sup>4)5)</sup> and so the details are omitted.

Fig. 1 shows the cross-section of the electrodes. Fig. 1 (a) is the case of single-conductor and Fig. 1 (b) the case of double-conductor, respectively.



(a) Single-Conductor (b) Double-Conductor  
Fig. 1 Cross-Section of Electrodes

A X-Y Recorder was used for measuring the corona starting voltage. The total corona current  $I_t$  and the applied voltage  $V$  were recorded on the Y- and X-axes, respectively. The corona starting voltage is defined as the value corresponding to the abrupt increase point of  $I_t$ .

A X-T Recorder was used for measuring the distribution of the d.c. corona current. The terminal voltage-drop across the standard resistance means the magnitude of the corona current.

The authors define the directivity factor  $\kappa$  in order to analyse the corona current

distribution quantitatively as shown in the previous papers;<sup>4)5)</sup> that is,

$$\text{Directivity Factor } \kappa = \frac{I_{\max}}{I_{\min}} = \frac{I_{90^\circ \text{ or } 270^\circ}}{I_{0^\circ \text{ or } 180^\circ}} \dots\dots(1)$$

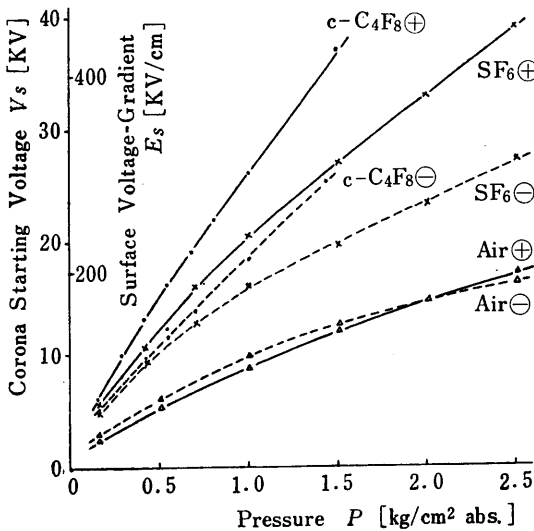
**3. Corona Starting Voltages in Various Gases**

Fig. 2(a) shows the corona starting voltages in air, SF<sub>6</sub>, c-C<sub>4</sub>F<sub>8</sub>, C<sub>2</sub>F<sub>6</sub> and N<sub>2</sub>O around the single-conductor and Fig. 2(b) those around the double-conductor, respectively. The voltage-gradient of the corona starting at the surface of the conductors, E<sub>s</sub>, is also shown on the ordinate axis. E<sub>s</sub>, at the applied voltage of V(kV), is given as follows;

Single-Conductor.....  
 $E_s = 11.5 \times V(\text{kV}) \dots\dots(2)$

Double-Conductor.....  
 $E_s = 8.49 \times V(\text{kV}) \dots\dots(3)$

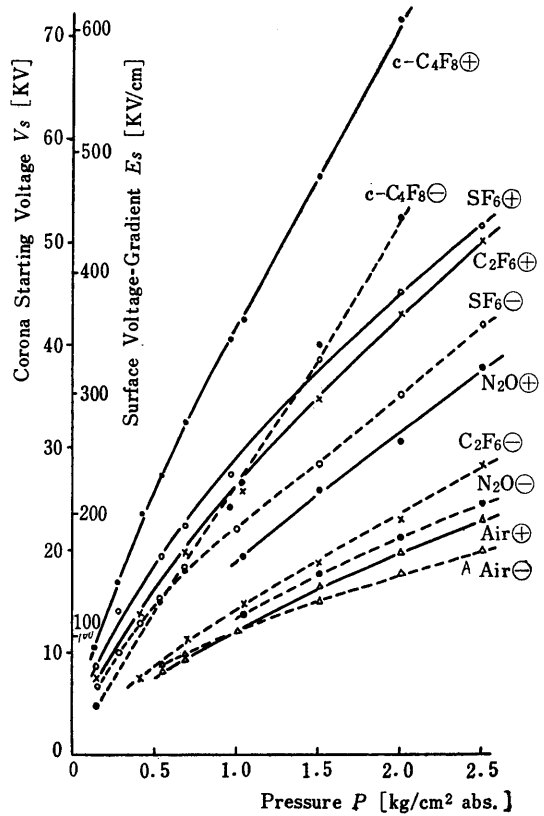
When the maximum voltage-gradient of the double-conductor is compared with that of the single-conductor, they are coincident within the limits of ±5%. The corona starting voltage of the double-conductor increases 30~40% in c-C<sub>4</sub>F<sub>8</sub> and SF<sub>6</sub>.



⊕ and ⊖ mean the potentials to be positive and negative, respectively.

**Fig. 2(a)** Corona Starting Voltages in Various Gases around Single-Conductor

Fig. 2(b) shows the corona starting voltages (abbrev. V<sub>s</sub>) around the double-conductor in various gases. V<sub>s</sub> of c-C<sub>4</sub>F<sub>8</sub> ⊕ is the highest



**Fig. 2(b)** Corona Starting Voltages in Various Gases around Double-Conductor, m/d = 10

of all the gases used in overall pressure. In less than 1 atm, however, V<sub>s</sub> of c-C<sub>4</sub>F<sub>8</sub> ⊖ is lower than that of SF<sub>6</sub> ⊕. The differences between positive and negative V<sub>s</sub> is nearly constant in more than 1 atm in both case of c-C<sub>4</sub>F<sub>8</sub> and SF<sub>6</sub>. The difference in c-C<sub>4</sub>F<sub>8</sub> is ca. 20 kV and in SF<sub>6</sub> ca. 10kV. V<sub>s</sub> of C<sub>2</sub>F<sub>6</sub> ⊕ is 2~4 kV lower than that of SF<sub>6</sub> ⊕ in overall pressure and V<sub>s</sub> of C<sub>2</sub>F<sub>6</sub> ⊖ is ca. 10 kV lower than that of SF<sub>6</sub> ⊖. So that, from the viewpoint of the electrical insulation, C<sub>2</sub>F<sub>6</sub> is not so useful as c-C<sub>4</sub>F<sub>8</sub>. However, it has been advocated to mix C<sub>2</sub>F<sub>6</sub> with c-C<sub>4</sub>F<sub>8</sub> in order to lower the high boiling point of c-C<sub>4</sub>F<sub>8</sub>.

In Fig. 2(b), is also shown V<sub>s</sub> of N<sub>2</sub>O, which is mixed with perfluorocarbon or SF<sub>6</sub>. V<sub>s</sub> of N<sub>2</sub> ⊕ is 3~4 kV lower than that of SF<sub>6</sub> ⊖ in more than 1 atm. V<sub>s</sub> of N<sub>2</sub>O ⊖ is a little higher than that of air ⊕.

In Fig. 2(b), the characteristics curves of the corona starting voltages of single-conductor show the same tendency as those of double-conductor.

**Table. I** Calculation of the Constants A and C

A and C		A		C	
Gases	Polarity	Positive	Negative	Positive	Negative
SF <sub>6</sub>		67.58	76.89	0.3054	0.1422
c-C <sub>4</sub> F <sub>8</sub>		72.8	118.7	0.375	0.0898

Table. I shows the constants A and C when Watson's equation (4) is applied to c-C<sub>4</sub>F<sub>8</sub> and SF<sub>6</sub>. A and C are calculated in consideration of the two cases of *r*, that is *r*=0.005 cm and *r*=0.015 cm.

Watson's Equation

$$E_s = A m_1 \delta \left( 1 + \frac{C}{\sqrt{\delta r}} \right) \dots\dots\dots(4)$$

where

*E<sub>s</sub>* : corona starting voltage-gradient in kV/cm

*m<sub>1</sub>* : coefficient of surface condition

*δ* : relative air density

*r* : radius of curvature of the conductor in cm

A&C : constants determined by the polarity and gases

**4. Directivity Characteristics in Various Gases**

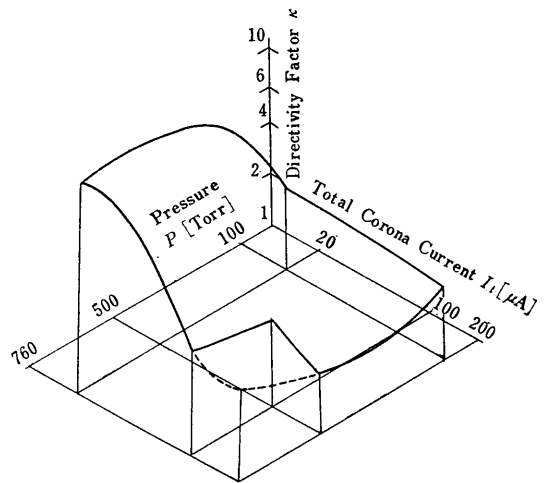
The authors have reported the directivity characteristics of SF<sub>6</sub>, c-C<sub>4</sub>F<sub>8</sub>, CO<sub>2</sub> and N<sub>2</sub> in detail in the 5 th paper.<sup>(5)</sup> In this section we investigate the directivity characteristics of C<sub>2</sub>F<sub>6</sub> (which is one of the perfluorocarbons, like c-C<sub>4</sub>F<sub>8</sub>) and N<sub>2</sub>O.

4-1. Directivity Factor in C<sub>2</sub>F<sub>6</sub> Gas

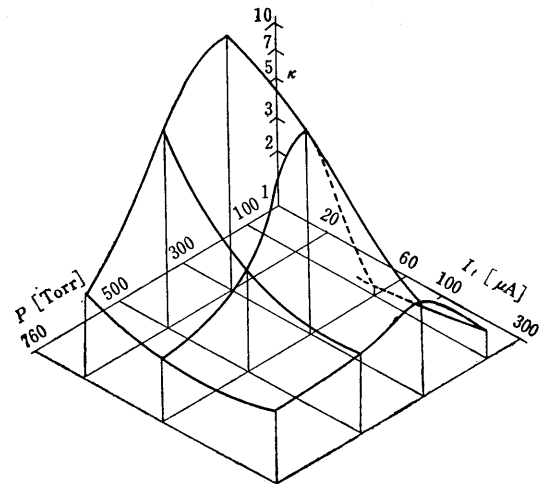
Fig. 3 shows the directivity characteristics of C<sub>2</sub>F<sub>6</sub>. The *κ*-*I<sub>t</sub>* characteristics of C<sub>2</sub>F<sub>6</sub> cannot be always represented by the equation of *κI<sub>t</sub><sup>β</sup>*=A, like in the case of c-C<sub>4</sub>F<sub>8</sub>.

In Fig.3 (a), the dependency of *κ* on pressure *p* in C<sub>2</sub>F<sub>6</sub> ⊕ is the smallest of all gases used and that of *κ* on the total corona current *I<sub>t</sub>* in C<sub>2</sub>F<sub>6</sub> is smaller than in c-C<sub>4</sub>F<sub>8</sub>. The directivity factor *κ* in the region of the lower pressure and the smaller corona current, could not be obtained because the distribution of the corona current was unstable.

In Fig. 3 (b), *κ* of C<sub>2</sub>F<sub>6</sub> ⊖ has the different characteristics from that of C<sub>2</sub>F<sub>6</sub> ⊕. That is to say, the dependency of *κ* on pressure is small in the region of the large *I<sub>t</sub>* and the maximum point of *κ* arises at *p*=ca. 300 Torr



(a) Positive Corona



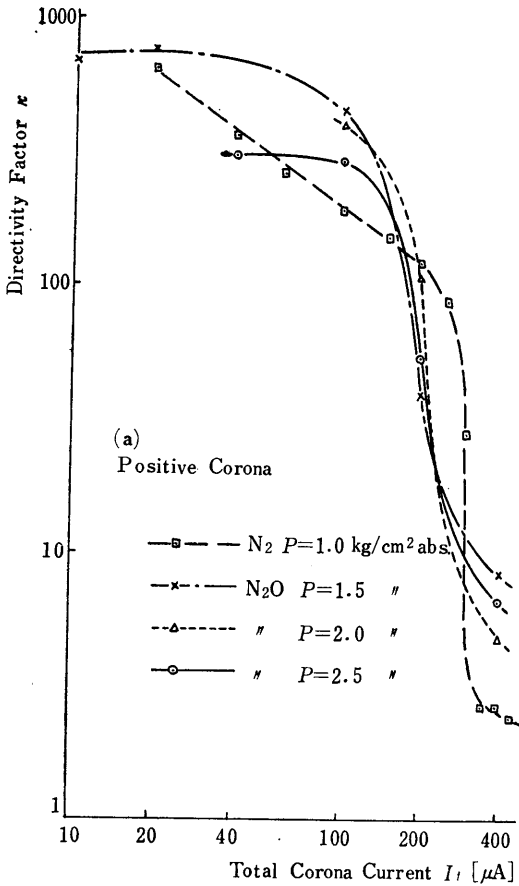
(b) Negative Corona

**Fig. 3** Directivity Characteristics of Corona Current in C<sub>2</sub>F<sub>6</sub>

in the region of the small *I<sub>t</sub>*. These characteristics are the similar as observed in SF<sub>6</sub> ⊖<sup>(5)</sup>.

4-2. Directivity Factor in N<sub>2</sub>O Gas

Fig. 4 shows the directivity characteristics of N<sub>2</sub>O. *κ*-*I<sub>t</sub>* characteristics of the positive corona in N<sub>2</sub>O are shown in Fig. 4 (a) and *κ*



(a) Positive Corona

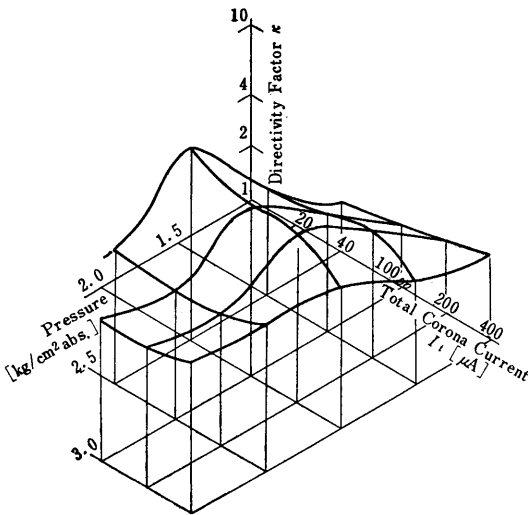


Fig. 4 Directivity Characteristics of Corona Current in N<sub>2</sub>O

of N<sub>2</sub>⊕ is also shown. In this figure, κ-I<sub>t</sub> characteristics of N<sub>2</sub>O⊕ are similar to those

of N<sub>2</sub>⊕. That is, κ-I<sub>t</sub> characteristic curves in N<sub>2</sub>⊕ at p=1kg/cm<sup>2</sup> abs. are separated into two regions at I<sub>t</sub>=ca. 300 μA.<sup>4)</sup> Similarly, in N<sub>2</sub>O there are two regions at ca. 200 μA.

On the contrary in the case of the negative corona, there are the differences between N<sub>2</sub>⊖ and N<sub>2</sub>O⊖. κ in N<sub>2</sub>⊖ could not be accurately obtained because of the unstable distribution of the corona current. In N<sub>2</sub>O⊖, however, the corona current distribution is stable and the κ-I<sub>t</sub>-p characteristics are shown in Fig. 4 (b).

The differences of the distribution between N<sub>2</sub>⊖ and N<sub>2</sub>O⊖ are explained by the existence of the electron attachment effect in electronegative gas (N<sub>2</sub>O).

### 5. Corona Starting Voltages in Various Gases Mixing with N<sub>2</sub>O

Fig. 5 shows the corona starting voltages V<sub>s</sub> in various gases mixing with N<sub>2</sub>O. The volume percentage of N<sub>2</sub>O (abbrev. N<sub>2</sub>Ov.%) is indicated on the quadrature axis. It is

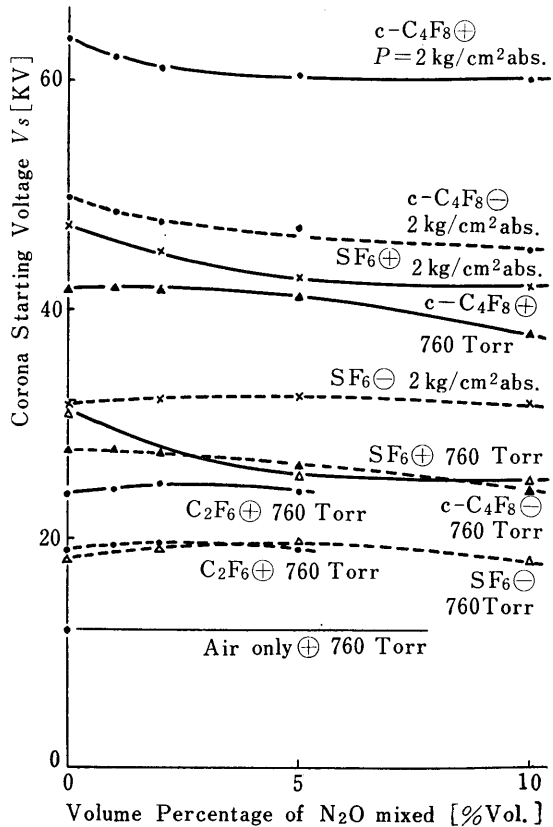


Fig. 5 Corona Starting Voltages in Various Gases Mixing with N<sub>2</sub>O Gas

noted that  $V_s$  of  $C_2F_6 \oplus$  mixing with  $N_2O$  increases a little (ca. 2 kV) at  $p_{total}=760$  Torr when  $N_2O$  v. % is about 2%. The same characteristics are observed in  $C_2F_6 \ominus$ . When  $N_2O$  is mixed with  $c-C_4F_8$ ,  $V_s$  decreases gradually as increment of  $N_2O$  v. % and  $V_s$  at  $N_2O$  v. % = 10% is ca. 4 kV lower than that of the pure  $c-C_4F_8$ .

It is worth considering to mix  $N_2O$  with perfluorocarbon in application to electrical insulation, because  $N_2O$  is effective to prevent carbon accumulation. It must be paid attention that  $N_2O$  gas is anesthetic and the chemical property of  $N_2O$  is similar to that of  $O_2$ .

In the characteristic curve of  $V_s$  in  $SF_6 \ominus$  mixing with  $N_2O$ , the maximum point arises at  $N_2O$  v. % = 5~10% at both  $p=760$  Torr and  $2 \text{ kg/cm}^2$  abs.

### 6. Directivity Characteristics in Various Gases Mixing with $N_2O$

Fig. 6 shows the comparison of the positive corona current distribution between  $c-C_4F_8$  and  $c-C_4F_8$  mixing with  $N_2O$  (percentage of mixing 95 : 5% vol.) under the condition of  $I_t=100 \mu A$  and  $p_{total}=760$  Torr. In Fig. 6, it is difficult to distinct the differences between both the corona current distribution. The corona current in the direction of  $\theta=0^\circ$  and  $180^\circ$ , if anything, increases a little in the mixed gas more than that in the pure gas.

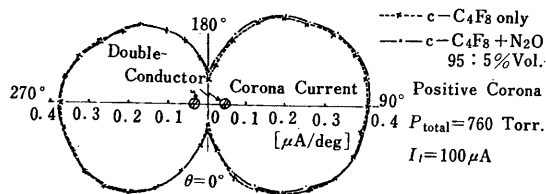
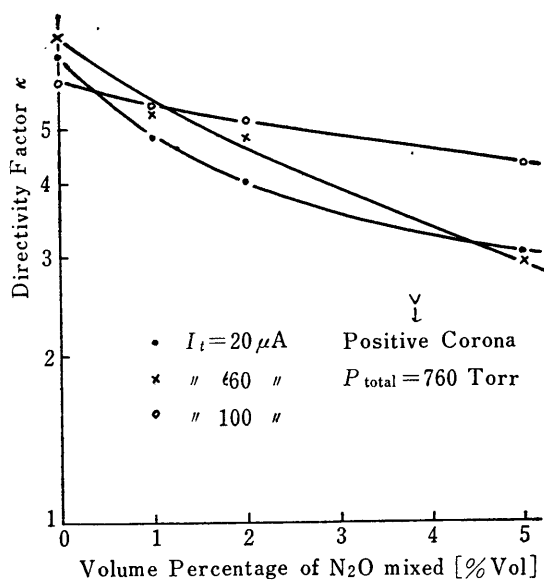


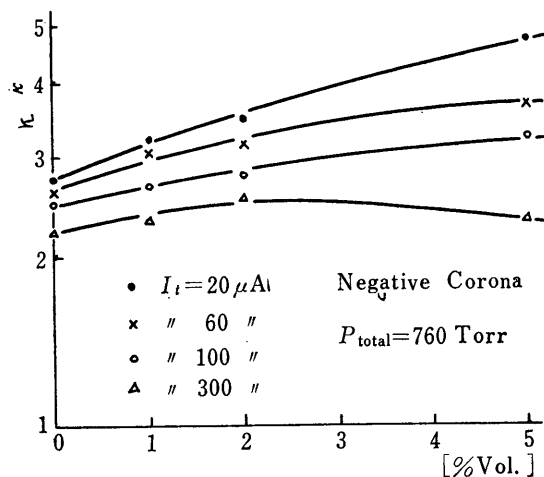
Fig. 6 An Example of Corona Current Distribution in  $c-C_4F_8$  mixing with  $N_2O$

Fig. 7 and Fig. 8 show the directivity characteristics in perfluorocarbon mixing with  $N_2O$ , 0~5% vol.  $\kappa$ -axis is graduated in logarithms scale.

Fig. 7(a) shows the directivity characteristics of the positive corona current distribution in  $c-C_4F_8$  mixing with  $N_2O$  at  $p_{total}=760$  Torr.  $\kappa$  tends to decrease as increment of  $N_2O$  v. %. On the contrary, as shown in Fig. 7(b),  $\kappa$  of the negative corona characteristics



(a) Positive Corona



(b) Negative Corona

Fig. 7 Directivity Characteristics of Corona Current in  $c-C_4F_8$  mixing with  $N_2O$

increases as increment of  $N_2O$  v. %. The rate of the decrease (or increase) of  $\kappa$  gets small as  $I_t$  increases.

Fig. 8 shows the directivity characteristics of the negative corona current distribution in  $C_2F_6$  mixing with  $N_2O$  at  $p_{total}=760$  Torr. In the  $\kappa$ -characteristic curve, a maximum point appears where  $N_2O$  v. % is ca. 2%.

Even if  $N_2O$  is mixed with perfluorocarbon below ca. 5%, the changes of the corona current distribution are not always large.

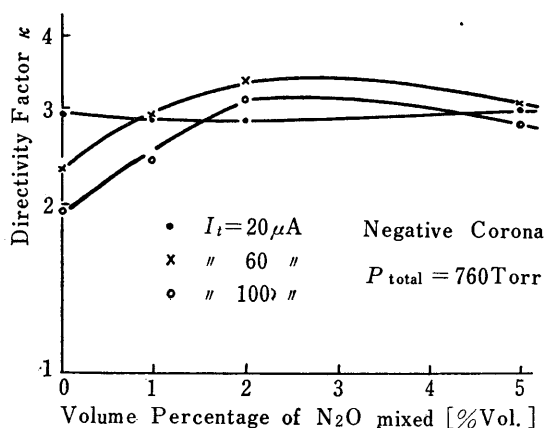


Fig. 8 Directivity Characteristics of Corona Current in  $C_2F_6$  mixing with  $N_2O$

### 7. Gas Analysis of Decomposed Products in Perfluorocarbon Exposed to Corona Discharge

$c-C_4F_8$  at ordinary state is not toxic for

human body and the thermal stability is very excellent<sup>7)</sup>, while are unknown the kinds and amounts of the products which are decomposed owing to the injection of electrical energy like corona discharge. The authors were confronted with the facts that  $c-C_4F_8$  exposed to corona discharge would produce the decomposed poisonous materials. Therefore, we made an attempt to analyse  $c-C_4F_8$  exposed to corona discharge by means of gaschromatograph.

The relation between the condition of the corona discharge, the kinds of the sampled gases and the purity are shown in Table. 2. This analysis is enough sufficient, because the rated purity of  $c-C_4F_8$  is 99.5 %.

The important results of the gaschromatographic analysis are as follows;

- (1)  $CF_3H$  was the most of all the decomposed products.

After corona discharge,  $CF_3H$  in sam-

Table. 2 Condition of Sampled Gases

Sampled Gases		Condition of Corona Discharge				Purity (%)
		Polarity	Total Corona Current	Hours	Pressure	
No. 1	$c-C_4F_8$ only	Negative	200 $\mu A$	2 hr.	1.4 kg/cm <sup>2</sup> abs.	98.0104* (98.4549)
No. 2	$c-C_4F_8 + N_2O$ 95 : 5 % Vol.	Negative	200 $\mu A$	2 hr.	1.2 kg/cm <sup>2</sup> abs.	98.3452* (98.7802)
No. 3	Pure $c-C_4F_8$	not exposed to Corona Discharge				99.5246* (99.9716)

\* means the purity of the gases taken off  $H_2O$ .

pled gas No. 1 increased 7500 times more than that in original gas and  $CF_3H$  in No.2 5100 times, respectively. The purity of  $CF_3H$  in original gas is 0.0002 %, but after corona discharge that became 1.4960 % in sampled gas No. 1 and 1.0220 % in No. 2, respectively.

- (2) KC-216 which was the poisonous materials was not detected.
- (3) Another fluorides were little detected. (In consideration of the sensitivity of gaschromatograph, the rest amounts are at least less than 0.0001 %.)
- (4)  $CO_2$  and  $H_2O$  hardly increased even

after the corona discharge.

At present the authors carry on the investigations of the gaschromatographic analysis in more detail.

### 8. Conclusions

- (1) It is pointed out that the  $\kappa-I_t$  characteristics of  $C_2F_6$  cannot be always represented by the equation of  $\kappa I_t^B = A$ , like in the case of  $c-C_4F_8$ .
- (2) The  $\kappa-I_t$  characteristics of  $N_2O^+$  are similar to those of  $N_2^+$ . On the contrary, in the characteristics of the negative corona, there are remarkable differences between  $N_2O^-$  and  $N_2^-$ .

- (3) The corona starting voltage  $V_s$  of  $C_2F_6$  mixing with  $N_2O$  increases a little when the volume percentage of  $N_2O$  is about 2%. And in the characteristic curve of  $V_s$  in  $SF_6 \ominus$  mixing with  $N_2O$ , a maximum point arises at  $N_2O$  v. % = 5~10%.
- (4) It is worth considering to mix  $N_2O$  with perfluorocarbon in application to electrical insulation. Because  $N_2O$  is effective to prevent carbon accumulation on the electrodes.
- (5) Even if  $N_2O$  is mixed with perfluorocarbon below ca. 5%, the changes of the corona current distribution are not always large.
- (6) The results of the gaschromatographic analysis gives that  $c-C_4F_8$  is decomposed by the corona discharge.  $CF_3H$  is the most of all the decomposed products.

### Acknowledgements

The authors wish to thank the Mitsui-Fluorochemical Co. for carrying out the gas analysis and for supplying  $c-C_4F_8$  and  $C_2F_6$  and the information of perfluorocarbon to us, and Mr. T. Shibata, Mr. H. Yoshii and Mr. T. Suzuki for their help and cooperation.

### References

- (1) S. Muto: Bulletin of Nagoya Institute of Technology vol. **15** pp. 253~258 (1963)
- (2) S. Muto: *ibid.* vol. **16** pp. 264~268 (1964)
- (3) S. Muto & T. Tani: *ibid.* vol. **17** pp. 246~254 (1965)
- (4) S. Muto, T. Inaba & T. Shibata: *ibid.* vol. **19** pp. 283~294 (1967)
- (5) S. Muto, T. Shibata & M. Udaka: *ibid.* vol. **20** pp. 257~263 (1968)
- (6) C. M. Brock, et al: Technical Bulletin of Du Pont (EL-10)
- (7) *ibid.* (EL-1)