# Origin of large thermoelectric power in off-stoichiometric Fe<sub>2</sub>VAl-based alloys

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**Abstract**. The valence-band and core-level soft x-ray photoelectron spectroscopy has been applied to investigate the origin of the large thermoelectric power in the off-stoichiometric and partially-substituted Fe-based Heusler-type alloys  $Fe_{2-x-y}Ir_yV_{1+x}Al$  and  $Fe_{2-x}V_{1+x-y}Ti_yAl$ . The non-rigid-band-like change in the valence-band electronic structure near the Fermi level  $E_F$  is found for the off-stoichiometric change x in Fe and V concentrations, whereas the rigid-band-like shift of  $E_F$  is recognized on the Ir or Ti substitution y. The non-rigid-band-like change may be induced by the excess-V or Fe (anti-site defects)-derived states in the pseudogap of Fe<sub>2</sub>VAl, while the rigid-band-like shift may be caused by a common band formation, where the d states of the substituted elements are incorporated into the main Fe-V 3d bands. The V2p core-level spectra consistently show the satellite structure due to the excess V as the anti-site defect. A model electronic structure is presented for explaining the observed enhancement of their thermoelectric power.

# 1. Introduction

Recently it has been found that the thermoelectric power is remarkably enhanced by partial substitution of Ir for Fe (or of Ti for V) together with off-stoichiometric change of Fe/V concentration ratio; the thermoelectric power becomes -0.17 mVK<sup>-1</sup> in the n-type Ir-substituted off-stoichiometric alloys Fe<sub>2-x-y</sub>Ir<sub>y</sub>V<sub>1+x</sub>Al [1] and to +0.11 mVK<sup>-1</sup> in the p-type Ti-substituted ones Fe<sub>2-x</sub>V<sub>1+x-y</sub>Ti<sub>y</sub>Al [2], respectively. Theoretical and experimental studies [3, 4] show that an intermetallic compound Fe<sub>2</sub>VAl possesses a sharp pseudogap across the Fermi energy  $E_F$ , which may lead to the enhancement of the thermoelectric power by changing  $E_F$  position on partial substitution of the forth element with more or less valence electrons than the original constituent elements without altering the valence-band electronic structure (so-called rigid band model) [5], because the thermoelectric power S(T) at temperature T may depend on the electronic density of states N(E) as follows [6];

$$S(T) \simeq \frac{1}{eT} \frac{\int N(E)(E - E_{\rm F}) \frac{\partial f(E;T)}{\partial E} dE}{\int N(E) \frac{\partial f(E;T)}{\partial E} dE}.$$
(1)

Here, f(E;T) stands for the Fermi-Dirac distribution function at T. Indeed thermoelectric powers of the substituted alloys are on a universal curve when they are plotted as a function of averaged valence

electron counts per constituent atom (VEC) [5], which can be semi-quantitatively explained by equation (1) [4,6]. However, the off-stoichiometric Fe/V change results in quite different dependence of the thermoelectric power on VEC in a non-rigid-band-like manner and its further enhancement [1,2].

In this study, we will present model electronic structures to clarify the origin of the remarkable enhancement of thermoelectric power in the partially substituted alloys with the off-stoichiometric Fe/V ratio, based on recent valence-band and core-level photoelectron measurements, which have been partly reported elsewhere [2,7]. According to equation (1), thermoelectric power is determined by N(E) around  $E_F$  within the energy range of ~4 $k_BT$ , where  $k_B$  is the Boltzmann constant, and its sign becomes positive (p-type) and negative (n-type) for large and small contributions, respectively, of the density of valence-band states  $N(E < E_F)$  compared to the conduction-band one  $N(E > E_F)$ .

# 2. Experimental

Photoelectron measurements were carried out at BL27SU of SPring-8, Japan Synchrotron Radiation Research Institute (JASRI). Linearly polarized and monochromatized soft x-rays from an undulator were used as an excitation photon source and photoelectrons from samples cooled down to ~10 K were analyzed in the angle-integrated mode with a hemispherical analyzer [8]. Total energy resolution was estimated as 0.16 eV at the excitation photon energy of ~1000 eV. Samples were polycrystalline of Fe<sub>2-x-y</sub>Ir<sub>y</sub>V<sub>1+x</sub>Al and Fe<sub>2-x</sub>V<sub>1+x-y</sub>Ti<sub>y</sub>Al and their clean surfaces for the photoelectron measurement were prepared by in situ fracturing. Details of the measurement and sample preparation are described elsewhere [1,2,7].

#### 3. Results and discussion

Figure 1 compares typical valence-band spectra of some of studied alloys. The stoichiometric alloy Fe<sub>2</sub>VAI with small but positive thermoelectric power shows a small shoulder structure at  $E_F$  in the photoelectron spectrum, indicating the pseudogap predicted for the alloy as having a semimetallic band structure [6]. A small shoulder structure around  $E_F$ , indicative of the pseudogap, is also recognized for the n-type thermoelectric alloys with the excess V (x > 0). This shoulder structure for the n-type alloys suggests that the conduction-band density of states may be increased just above  $E_F$ , although unoccupied electronic states cannot be directly proved by photoelectron spectroscopy. On the other hand, a monotonic steep increase of the main d band is clearly seen from  $E_F$  to  $E_B \sim 0.3$  eV for the p-type alloys with the excess Fe (x < 0), showing the valence-band edge which may lead to the



**Figure 1.** Valence-band electronic structures of  $Fe_2VAI$ -based off-stoichiometric alloys. Model electronic structures are shown in the bottom and top for anti-site V and Fe defects, respectively, where majority (down) spin 3*d* states of V and Fe appear below the Fermi energy  $E_F$  (thin vertical lines) in a semimetallic pseudogap with Fe3*d* valence band (Fe3*d* VB) and V3*d* conduction band (V3*d* CB), and unoccupied minority (up) spin V3*d* states are located just above  $E_F$ . Thick broken lines are drawn as a guide for eyes.

positive thermoelectric power. Thus the spectral findings confirm the relation between the thermoelectric power and the density of states.

In Fe<sub>2-x</sub> $V_{1+x}$ Al, the off-stoichiometric change in the Fe/V ratio causes no rigid-band-like shift of the valence-band structures. Instead, a band at the binding energy  $E_{\rm B}$  of about 0.5 eV is enhanced for the excess V concentration x > 0, and a new band appears at  $E_{\rm B} \sim 0.3$  eV for the excess Fe concentration x < 0. These new bands can be attributed to the anti-site V and Fe defects, which may be formed around the excess V and Fe occupying at the Fe and V sites, respectively, in the Heusler( $L_2_1$ )-type Fe<sub>2</sub>VAl lattice structure. According to Bansil et al. [9], 3d states of the transition-metal elements in the octahedral site (the V site in Fe<sub>2</sub>VAl) form a pair of bonding and anti-bonding bands for each spin subbands, while those in the tetrahedral site (the Fe site in  $Fe_2VAI$ ) show a triple-band structure consisting of bonding and anti-bonding bands and an extra band between them. For the anti-site Fe defect, its body-centered-cubic-like local cluster brings such anti-bonding majority (down) spin Fe3d states below  $E_F$  [9], as shown in the top of figure 1. This may correspond to the 0.3-eV band observed for the Fe-rich (x < 0) alloys. On the other hand, spin-projected densities of states calculated for V<sub>2</sub>FeAl [10] suggests that the anti-site V in the tetrahedral site may induce a majority (down) spin V3d band below  $E_{\rm F}$  and a minority (up) spin V3d band above  $E_{\rm F}$  in the pseudogap, as drawn in the bottom of figure 1. This is supported by a band structure calculation with a supercell method for  $Fe_{2-x}V_{1+x}Al$  [11]. The 0.5eV band for the V-rich (x > 0) alloys may be attributed to such a majority spin V3d state.

In other point of view, the excess Fe and V might induce so-called impurity states in the pseudogap, which is brought about by small overlap between a mainly Fe3*d*-derived valence-band top and almost purely V3*d*-derived conduction-band bottom [3,12]. Since Fe(II) is more electronegative than Fe(I) in  $D0_3$ -ordered Fe<sub>3</sub>Al [9,13], which can be regarded as  $L2_1$ -type Fe(II)<sub>2</sub>Fe(I)Al, the anti-site Fe, similar to Fe(I), may have less valence electrons than Fe at the Fe(II) site and the anti-site V may increase its valence electrons. Indeed, the V2*p* core-level spectra show new chemically shifted components in the low binding energy side of the main line for the excess V concentration x > 0 [7], implying the increase in the local valence electrons of the anti-site V. Although satellite structures were not observed for the Fe and Al 2*p* core-levels at present, the anti-site Fe and V might result in Fe and V 3*d* states split off from the valence and conduction bands, respectively. Thus 0.3-eV band appearing in the Fe-rich alloys might be ascribed to such split-off Fe 3*d* states. For the V-rich alloys, the anti-site V forms a V cluster with neighboring V tetrahedrally coordinated and hence the bonding and anti-bonding states between them might appear below and above  $E_F$ , respectively. This might cause the 0.5 eV band in the V-rich alloys.

Further Ir and Ti substitutions lead to rigid-band-like shifts of the first and second peaks observed respectively at  $E_{\rm B} \sim 0.8$  and 1.6 eV, where thick broken lines are drawn as a guide for eyes in figure 1; the peaks are shifted towards the low and high binding energy sides for Ti and Ir substitutions, respectively. This rigid-band-like behavior can be explained in terms of a common band formation [9,14] between the Fe-V 3d bands and the substituted Ir 5d or Ti 3d states. The d states of Ir and V occupying at the Fe and V sites may be moderately incorporated into the main Fe-V 3d bands. Close inspection of the intensity at  $E_{\rm F}$  indicates its increase and decrease on the Ti and Ir substitutions, respectively. Taking account of the experimental energy resolution of 0.16 eV and sample temperature of ~10 K, these changes in the intensity at  $E_{\rm F}$  suggest further increasing difference between the densities of states below and above  $E_{\rm F}$ ,  $N(E < E_{\rm F})$  and  $N(E > E_{\rm F})$ , on substitution, which implies further enhancement of their thermoelectric powers in Fe<sub>2-x</sub>yIr<sub>y</sub>V<sub>1+x</sub>Al and Fe<sub>2-x</sub>V<sub>1+x-y</sub>Ti<sub>y</sub>Al.

Thus the present results show that the thermoelectric power in those alloys is enhanced by the non-rigid-band-like drastic change caused by the anti-site defects and the rigid-band-like moderate shift by the common band formation between the Fe-V 3d bands and the substituted Ir 5d or Ti 3d states under the presence of the pseudogap.

### 4. Conclusion

We have investigated the origin of the large thermoelectric power in the off-stoichiometric and partially-substituted Fe-based Heusler-type alloys  $Fe_{2-x-y}Ir_yV_{1+x}Al$  and  $Fe_{2-x}V_{1+x-y}Ti_yAl$  by soft x-ray

photoelectron spectroscopy. We have found the non-rigid-band-like change in the valence-band electronic structure near  $E_F$  for the off-stoichiometric change x in Fe and V concentrations, the rigid-band-like shift of  $E_F$  on the Ir or Ti substitution y, and the satellite structure due to the anti-site V in the V2p core-level spectra. The non-rigid-band-like change may be induced by the anti-site defects-derived states in the pseudogap of Fe<sub>2</sub>VAl, while the rigid-band-like shift is caused by a common band formation, where the d states of the substituted elements are incorporated into the main Fe-V 3d bands. We have proposed model electronic structures to explain the observed enhancement of the thermoelectric power, based on our experimental results and so far reported theoretical studies; the positive enhancement of thermoelectric power in the Fe-rich alloys is attributed to the increase in the density of anti-site Fe-derived majority spin states below  $E_F$  and the negative enhancement in the V-rich alloys to the increase in the density of unoccupied anti-site V-derived minority spin states above  $E_F$ . Assumed unoccupied states will be clarified by other methods such as inverse photoelectron spectroscopy.

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