Instrumental Achievements

– X-Ray Analysis ——

## Crystal Structure of Diquinolyl-8,8'-disulfide

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8-Quinolinethiol, a selective chelating reagent towards soft metal ions<sup>1</sup>, is usually stored in its oxidized form, diquinolyl-8,8'-disulfide, for easy handling. In this study, we carried out an X-ray structural analysis for this compound. The molecule adopts an approximate  $C_2$ symmetry: the atoms in one of the halves, quinolyl-8sulfide, are represented by the primed numbers of the corresponding ones in the other. The H atoms were located on a difference Fourier map and refined with isotropic temperature factors set equal to the  $B_{eq}$  values of the bonded atoms. The corresponding bond

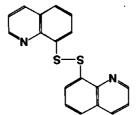


Fig. 1 Chemical structure.

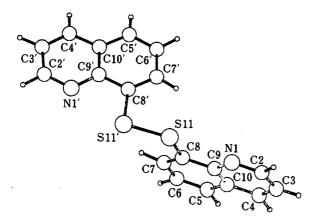


Fig. 2 Perspective view with atom-numbering.

distances and angles in the two halves agree within experimental error. The dihedral angle between the ring planes is  $78.8(1)^{\circ}$ .

Table 1 Crystal and experimental data

Formula: $C_{18}H_{12}N_2S_2$				
Formula weight=320.43				
Crystal system: monoclinic				
Space group: P2 <sub>1</sub> /c	Z=4			
a=13.602(2) Å				
b=9.540(1)	β=116.37(1)°			
c = 13.112(2)				
V=1524.4(3) Å <sup>3</sup>				
$D_x = 1.396 \text{ g/cm}^3$				
R=0.048				
No. of reflections used=213	No. of reflections used=2134			
Measurement: Rigaku AFC	Measurement: Rigaku AFC-5R			
Program system: private (XPACK86 SHIONOGI)				
Structure determination: MULTAN87				
Refinement:block-diagonal least-squares				

Table 2 Final coordinates (×10<sup>4</sup>) of non-H atoms

Atom	x	y	Z	$B_{ m eq}/10^2{ m \AA}^2$
N1	8310(2)	1913(3)	-297(2)	508(9)
C 2	8824(3)	2672(4)	-756(3)	659(15)
C 3	9969(3)	2741(4)	-331(3)	675(16)
C 4	10619(3)	2012(4)	620(3)	610(13)
C 5	10718(3)	357(4)	2132(3)	589(12)
C 6	10185(3)	-429(4)	2586(3)	597(12)
C 7	9037(3)	-451(3)	2112(3)	495(10)
C 8	8425(2)	334(3)	1162(2)	423(9)
69	8959(2)	1172(3)	659(2)	419(9)
C10	10119(2)	1177(3)	1149(3)	479(10)
S11	6966(1)	395(1)	454(1)	471(3)
N1'	5884(2)	-3429(3)	2100(2)	458(8)
C2'	5569(3)	-4477(3)	2537(3)	529(11)
C3'	5617(3)	-5891(3)	2275(3)	571(12)
C4 '	6008(2)	-6224(3)	1512(3)	519(11)
C5'	6768(2)	-5386(3)	200(3)	489(10)
C6'	7083(3)	-4284(3)	-243(3)	508(10)
C7'	7014(2)	-2890(3)	83(3)	475(10)
C8'	6617(2)	-2628(3)	855(2)	410(9)
69'	6268(2)	-3748(3)	1329(2)	385(8)
C10'	6354(2)	-5146(3)	1004(2)	422(8)
S11'	6480(1)	-937(1)	1359(1)	487(3)

 $B_{\rm eq} = (4/3) \Sigma_i \Sigma_j \beta_{ij} (\boldsymbol{a}_i \cdot \boldsymbol{a}_j).$ 

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Table 3 Bond distances (Å) and angles (°)					
N1 - C2	1.323(5)	N1 - C9	1.366(4)		
C2 - C3	1.403(6)	C3 - C4	1.357(6)		
C4 - C10	1.413(6)	C5 - C6	1.351(6)		
C5 - C10	1.416(6)	C6 - C7	1.401(6)		
C7 - C8	1.373(5)	C8 - C9	1.424(4)		
C8 - S11	1.780(3)	C9 - C10	1.415(5)		
N1' - C2'	1.315(5)	N1' - C9'	1.363(4)		
C2' - C3'	1.401(6)	C3' - C4'	1.363(6)		
C4' - C10'	1.415(5)	C5' - C6'	1.359(6)		
C5' - C10' C7' - C8' C8' - S11' S11 - S11' C2 -N1 -C9	1.417(5) 1.365(5) 1.784(3) 2.039(2) 116.4(3)	C6' - C7' C8' - C9' C9' - C10' N1 -C2 -C3	1.413(6) 1.420(4) 1.421(4)		
C2 -C3 -C4	119.7(4)	C3 -C4 -C10	118.7(4)		
C6 -C5 -C10	120.2(4)	C5 -C6 -C7	121.6(4)		
C6 -C7 -C8	120.1(4)	C7 -C8 -C9	119.9(3)		
C7 -C8 -S11	125.3(3)	C9 -C8 -S11	114.8(2)		
N1 -C9 -C8	117.4(3)	N1 -C9 -C10	123.4(3)		
C8 -C9 -C10	119.1(3)	C4 -C10 -C5	123.4(4)		
C4 -C10 -C9	117.5(3)	C5 -C10 -C9	119.1(3)		
C2'-N1' -C9'	117.4(3)	N1'-C2' -C3'	124.4(4)		
C2'-C3' -C4'	118.7(4)	C3'-C4' -C10'	119.7(3)		
C6'-C5' -C10'	119.9(3)	C5'-C6' -C7'	121.5(4)		
C6'-C7' -C8'	119.9(3)	C7'-C8' -C9'	120.4(3)		
C7'-C8' -S11'	125.5(3)	C9'-C8' -S11'	114.1(2)		
N1'-C9' -C8'	118.0(3)	N1'-C9' -C10'	122.8(3)		
C8'-C9' -C10' C4'-C10'-C9' C8 -S11 -S11'	119.2(3) 116.9(3)	C4'-C10'-C5' C5'-C10'-C9' C8'-S11'-S11	122.8(3) 123.9(3) 119.2(3) 103.5(2)		

## Reference

1. A. Yuchi, K. Sugiura, H. Wada and G. Nakagawa, *Bull. Chem. Soc. Jpn.*, 60, 4291 (1987) and references cited therein.

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