Structural Change in Phosphorus-Bearing Dicalcium Silicates

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リンを固溶したケイ酸カルシウムの構造変化

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Structures of $(Ca_{2-x/2}\Box_{x/2})$ $(Si_{1-x}P_x)O_4$ crystals were examined as a function of x ranging from 0.03 to 0.40. All of the samples were heated at the stable temperature region of the α -phase and then quenched in water. The phase constitution at ambient temperature was classified into three categories according to the fraction of the α -to- α'_{H} transition. When the transition was completed as in the crystals with $x \le 0.100$, the β - and α'_{L} -phases were obtained, the relative amounts of which were determined by the start and finish temperatures of the α'_{L} -to- β martensitic transformation. With $0.125 \le x \le 0.150$, the α -to- α'_{H} transition was incomplete. During further cooling, the product α'_{H} -phase was inverted to the α'_{L} -phase, and the residual α -phase was inverted to the incommensurate phase successively. Because the start temperature of the α'_{L} -to- β transformation was lower than ambient temperature, the α'_{L} -phase was stabilized. With $x \ge 0.175$, all of the crystals were free from the α -to- α'_{H} transition. The crystals with $0.175 \le x \le 0.225$ were made up exclusively of the incommensurate phase. A good correlation existed between the modulation wavelength (=N) and the P/(Si + P) ratio (=x) as $N=4.134-1.56 \times (0.175 \le x \le 0.250)$. With $0.275 \le x \le 0.300$, the crystals were isostructural with α -Ca₂SiO₄. The hexagonal phase with $0.350 \le x \le 0.400$, probably a transition product from the α -phase, showed two-dimensional modulations along the *a*-axis with N=2 and along the *c*-axis with N=3.

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1. Introduction

The phase diagram in the binary system Ca₂SiO₄-Ca₃(PO₄)₂¹⁾⁻⁵ shows that Ca₂SiO₄ (C₂S) incorporates certain amounts of phosphate to stabilize the high temperature modifications of β , α' and α with increasing phosphate content. When these Ca₂SiO₄ solid solutions (C₂S(ss)) were quenched from elevated temperatures, they, however, do not necessarily show exactly the same structures as the high temperature modifications. They sometimes show modulated distortions with incommensurate or commensurate superstructures.⁴),⁶-8

Incorporation of some kinds of foreign oxides into $C_2S(ss)$ can effectively depress the transition rate of $\alpha \rightarrow$ $\alpha'_{\rm H}$ and the transformation temperatures of $\alpha'_{\rm L} \rightarrow \beta$, leading to the stabilization of the high temperature modifications.⁹⁾⁻¹²⁾ The transition of $\alpha \rightarrow \alpha'_{\rm H}$ is an isothermal nucleation and growth process. The time and temperature for the start and finish of the transition have been represented by two C-shaped curves on the time-temperature-transformation (TTT) diagram.¹⁰ With increasing concentration of foreign oxides in solid solution, the time for the start of the transition and that for the finish increased steadily. By superimposing cooling curves on the TTT diagram, we can predict the intracrystalline microtexture as well as the constituent phases after cooling. The transformation of $\alpha'_{\rm L} \rightarrow \beta$ is martensitic.¹¹⁾⁻¹³⁾ The reaction spontaneously begins at a definite temperature, $M_{\rm s}$, and with decreasing temperature, the transformed fraction increases to $M_{\rm f}$, at which the reaction is completed. Because the transformation is athermal, the fraction depended entirely on temperature; the phase constitution at ambient temperature (T_a) was determined to be $\alpha'_{\rm L}$ $(M_{\rm s} < T_{\rm a})$, $\alpha'_{\rm L}$ and β $(M_{\rm f} \le T_{\rm a} \le M_{\rm s})$ and β $(T_{\rm a}$ $<\!M_{
m f}$).¹²⁾ With increasing concentration of foreign oxides, the transformation temperatures of M_s and M_f decreased

steadily.11),12)

An incommensurate orthorhombic phase has been obtained for the $C_2S(ss)$ quenched from the stable temperature region of the α -phase. $^{(4),6)-8),14)}$ In the reciprocal space, the reflections were expressed by

 $Q=ha^*+kb^*+lc^*+nk$ (1) using four indices (h, k, l and n), and the wave vector k was redefined by $(1/N)a^*+c^*$ with N as the modulation wavelength.⁸⁾ Commensurate modulations were reported for the hexagonal C₂S(ss) with P/(Si+P)=0.398.⁸⁾ They were N=2 along the *a*-axis and N=3 along the *c*-axis with reference to the underlying α -phase lattice.

The present study deals with the structural change with concentration of P_2O_5 in $C_2S(ss)$ obtained by quenching from the stable temperature region of the α -phase.

Experimental

2.1 Materials Fifteen kinds of mixtures were prepared from reagentgrade chemicals, CaCO₃, SiO₂ and CaHPO₄·2H₂O. They were pressed into pellets, heated at 1500°C for 5 days and then quenched in water. The chemical formulae of the crystals obtained after heating were (Ca_{2-x/2} $\square_{x/2}$) (Si_{1-x}P_x)O₄ with *x* ranging from 0.03 to 0.40 (Table 1). During heating, most of the crystal grains were developed larger than 50 μ m (<400 μ m) in diameter. The samples were termed S-A to S-O in the order of increasing *x*. Because these compositions varied along the line Ca₂SiO₄-Ca₃(PO₄)₂ on the binary phase diagram, they were also represented by (1-*y*) Ca₂SiO₄·*y* Ca₃(PO₄)₂, where y=x/(2-x) (0.015 $\leq y \leq 0.25$).

2.2 Characterization

Thin sections were made of all the samples, and the intracrystalline microtextures were observed under an optical microscope.

Crystal grains of approximately 100 μ m in diameter were selected for the Laue and precession method (samples S–J, S–K, S–M, S–N and S–O).

The powder X-ray diffraction (XRD) profiles were ob-

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		18	ible I.	Phase	Consti	tution	of the	$(\operatorname{Ca}_{2-x/})$	$2 \square_{x/2}$	$S_{1-x}P_x$	0_4 Cr	ystals			
Category	I		II			ш									
Sample	S-A	S-B	S-C	S-D	S-E	S-F	S-G	S-H	S-I	S-J	S-K	S-L	S-M	S-N	S-O
x	0.030	0.080	0.100	0.125	0.150	0.175	0.200	0.225	0.250	0.275	0.300	0.325	0.350	0.375	0.400
Phase	β	β	$\beta + \alpha'_L$	α'_{L} +INC	α' _L +INC	INC	INC	INC	INC+α	α	α	α + H	Н	Н	Н

INC: Incommensurate phase

H: Hexagonal phase with modulations N = 2 along the *a*-axis and N = 3 along the *c*-axis.

tained on a diffractometer (Philips Co., Model PW3050) using monochromatized Cu K α radiation (40 kV, 50 mA) and the step-scan technique (step width $= 0.02^{\circ}$ and fixed time = 10 s) in the 2θ range from 20 to 60°. Silicon powder was used as an internal standard (sample/Si=7 by weight). Peak positions were determined by fitting individual line profiles to the Pearson VII function on a computer program PRO-FIT.¹⁵⁾ With the crystals in S-A, S-B, S-J and S-K, the relative integrated intensities as well as the cell dimensions were refined by the whole-powder-pattern decomposition without reference to a structural model (WPPD) method.15)

X-ray powder diffraction intensities of the incommensurate phase were found from calculation on a simulation mode of the computer program RIETAN.¹⁶⁾ The crystallographic DATA used were those determined by Saalfeld and Klaska,⁶⁾ who, assuming N=4, refined the structure of $6Ca_2SiO_4 \cdot 1Ca_3(PO_4)_2$ (x=0.250 and y=0.143) with N= 3.75.

3. Results and discussion

3.1 Phase constitution with P/(Si+P) ratio

The phase constitution at ambient temperature (20°C) was classified into three categories (I, II and III) according to the fraction of the α -to- $\alpha'_{\rm H}$ transition (Table 1). In accordance with the previous results,¹⁰⁾ the time for the start of the α -to- $\alpha'_{\rm H}$ transition and that for the finish should increase steadily with increasing x. This leads to a systematic change in the transformed fraction when the crystals with various *x*-values were cooled at the same rate; the resulting phase constitution will be $\alpha'_{\rm H}$, $\alpha'_{\rm H} + \alpha$ and α with increasing x. These phases may be inverted to thermodynamically more favorable phases during further cooling.

Powder XRD and thin-section observation showed that the crystals in S–A and S–B ($x \le 0.080$) were composed exclusively of the β -phase and those in S-C (x=0.100) of both the α'_{L} - and β -phases (Table 1). The intracrystalline lamella structure $^{17),18)}$ within these crystals indicated that they were formed in the stable temperature region of the α phase. Upon quenching, the phase transition of $\alpha \rightarrow \alpha'_{\rm H}$ terminated and, during further cooling, the transitions of $\alpha'_{\rm H} \rightarrow$ $\alpha'_{\rm L} \rightarrow \beta$ occurred successively. The martensitic transformation of $\alpha'_{L} \rightarrow \beta$ was completed within the crystals in S-A and S-B; however, it was incomplete within the crystals in S-C because the $M_{\rm f}$ was below ambient temperature. Accordingly, the $M_{\rm f}$ decreased with increasing x as 20°C< $M_{\rm f}$ for $x \le 0.080$ and $M_{\rm f} < 20^{\circ}{\rm C}$ for x = 0.100.

The thin-section observation and precession photographs showed that the crystals in S–D and S–E $(0.125 \le x \le 0.150)$ consisted of the $\alpha'_{\rm L}$ -phase lamellae with the host incommensurate phase (Table 1). Both phases were twinned cyclically by 120° around the *c*-axis of the former α -phase.^{8),14)} This texture implied that the α -to- $\alpha'_{\rm H}$ transition was incomplete; the parent α -phase underwent the transition to form the

lamella structure, and the residual α -phase between the lamellae was subsequently inverted to the incommensurate phase. The $\alpha'_{\rm H}$ -phase was further inverted to the $\alpha'_{\rm L}$ -phase. Because the $M_{\rm s}$ was below ambient temperature, the $\alpha'_{\rm L}$ phase was stabilized. Thus, the compositional change of $M_{\rm s}$ was determined to be $20^{\circ}C < M_s$ for x = 0.100 (sample S-C) and $M_{\rm s} < 20^{\circ}{\rm C}$ for $x \ge 0.125$.

All the crystals with $0.175 \le x \le 0.400$ were free from the transition of $\alpha \rightarrow \alpha'_{\rm H}$ (Table 1). The crystals in S-F, S-G and S-H (0.175 $\leq x \leq 0.225$) were composed entirely of the incommensurate phase without the lamella structure, indicating that they were directly inverted from the α -phase. The crystals in S-J, S-K, S-L, S-M, S-N and S-O $(0.275 \le x \le 0.400)$ were uniform without twinning under crossed polars. The powder XRD pattern of the crystal in S-I (x=0.250) was characterized by overlapping of those of the crystals in S-H (x=0.225) and S-J (x=0.275). Thus, the crystal must be composed of the two phases.

The compositional changes of M_s and M_f were as a result determined to be $20^{\circ}C < M_f$ for $x \le 0.080$, $M_f < 20^{\circ}C < M_s$ for x=0.100 and $M_{\rm s}<20^{\circ}$ C for $x\geq0.125$. In martensitic transformations, deformation above $M_{s'}$, which may be achieved by quenching large crystal grains rapidly as in the present samples, can result in the formation of the product phase.¹⁹⁾ Thus, the $M_{\rm s}$ temperatures as determined may be higher than those at which the crystals were pulverized. In the previous studies, the Ms temperatures of the powdered samples were determined by DTA9),20) and high-temperature powder XRD.¹²⁾ A similar effect (i.e., grain-size effect) was observed for the tetragonal-to-monoclinic transition in ZrO₂ particles; the $M_{\rm s}$ was raised by increasing the particle size.^{21),22)}

3.2 Change in modulation wavelength of the incommensurate phase with P/(Si+P) ratio

The powder XRD intensities of the incommensurate phase found from calculation were in fair agreement with those observed. They are summarized in Table 2 together with the *d*-values.

The subcell dimensions of the crystals in S-F, S-G and S-H were determined from the reflections of 0220, 1120, 1300 and 2000 (Table 3). The diffraction profiles for 0220 and 1120 almost coincided with each other because the subcell was nearly orthonexagonal in geometry ($\sqrt{3} a_s/b_s \approx 1$).

The superstructure reflection with indices 1311 was relatively strong in intensity and overlapped no other reflections (Table 2). From Eq. (1), the reciprocal vector is given by

$$\dot{Q}_{13\bar{1}1} = (1+1/N)a_s^* + 3b_s^* \tag{2}$$

The magnitude is determined from the peak position $(2\theta_{13\overline{1}1})$ $|Q_{13\overline{1}1}| = 2\sin\theta_{13\overline{1}1}/\lambda$ (3)

where λ is the radiation wavelength. The $2\theta_{13\overline{1}1}$ -values were 35.367° (S-F), 35.443° (S-G) and 35.467° (S-H). Substitution of Eq. (2) into Eq. (3) and separation of N given

 $N = [a_{\rm s} \{ (2\sin\theta_{13\bar{1}1}/\lambda)^2 - (3/\bar{b}_{\rm s})^2 \}^{1/2} - 1]^{-1}$ (4)

Table 2. Powder X-ray Diffraction Intensities Found from Calculation for the Incommensurate Phase: $(Ca_{1.875} \square_{0.125}) (Si_{0.750} P_{0.250}) O_4 (30^\circ \le 2 \theta \le 36^\circ, Cu K\alpha_1)$

h	k	l	n	20 (°)	d _{calc} (nm)	Int.
2	1	1	-1	30.36	0.2942	<1
1	1	3	-1	30.45	0.2933	4
1	0	2	0	30.92	0.2889	17
1	2	0	1	30.94	0.2888	1
1	3	1	-1	31.08	0.2875	13
1	2	-2	2	31.20	0.2865	<1
0	3	1	0	31.38	0.2848	5
0	3	0	1	31.66	0.2824	2
2	0	2	-1	31.70	0.2821	3
0	2	2	0	32.38	0.2763	37
1	1	2	0	32.39	0.2762	86
1	3	3	-2	32.48	0.2754	2
0	2	1	1	32.65	0.2741	<1
1	3	0	0	32.99	0.2713	100
2	0	0	0	33.03	0.2710	55
2	1	2	-1	33.13	0.2702	2
0	2	0	2	33.44	0.2677	2
1	3	2	-1	33.80	0.2650	1
1	2	-1	2	33.91	0.2642	<1
2	1	0	0	34.41	0.2604	<1
2	2	1	-1	34.66	0.2586	2
1	2	3	-1	34.73	0.2581	4
1	1	1	1	34.74	0.2580	2
1	3	-1	1	35.31	0.2540	8
1	3	1	0	35.58	0.2521	<1

Crystallographic data given by Saalfeld and Klaska.⁶⁾

a = 0.542nm, b = 0.940nm, c = 0.683nm and N = 4.

The setting of axes differs from that given in the original article.

The modulation wavelengths determined by substitution in Eq. (4) of $a_{\rm s}$, $b_{\rm s}$ and $\theta_{13\bar{1}1}$ are given in Table 3. A good correlation existed between N and x (Fig. 1). In the figure, the open square and open circles indicate the previous results.^{6),8)} In the latter, the samples contained K₂O (0.12–0.25 mass%), Al₂O₃ (0.12–0.17 mass%) and Fe₂O₃ (0.63–0.69 mass%) in addition to P₂O₅. The incorporation of these oxides in solid solution may contribute to increasing the modulation wavelength, the increment of which is given by ΔN =0.236–0.94 x.

3.3 Structures of the hexagonal phases

The intensity distribution on the Laue and upper-level



Fig. 1. Modulation wavelength (N) of the incommensurate phase as a function of P/(Si+P) ratio. (\blacksquare) Present study (Samples S-F, S-G and S-H). (\bigcirc , \square) Previous studies.^{6),8)} Crystals represented by the open circles were those doped with K_2O , AI_2O_3 and Fe_2O_3 in addition to P_2O_5 .



Fig. 2. Precession photographs of the single crystal in S–K. Incident beam parallel to c (A) and a (B). Heavy lines depict reciprocal unit-cells. Zero level, Zr-filtered Mo K α , μ =30°.

Table 3. Crystallographic Data for $C_2S(ss)$

Sample	Crystal System	Modulation	Subcell Dimensions						
oumple	Crystal Bystelli	Woddiation	a _s (nm)	b _s (nm)	c _s (nm)	β (°)	Volume (nm ³)		
S-A	Monoclinic	None	0.55075(1)	0.67550(5)	0.93132(4)	94.273(4)	0.34552(5)		
S-B	Monoclinic	None	0.55041(1)	0.67604(5)	0.93191(4)	93.994(4)	0.34592(5)		
S-F	Orthorhombic	3.87a	0.5439(2)	0.9394(6)	0.6828(4)		0.3489(3)		
S-G	Orthorhombic	3.81 <i>a</i>	0.5435(3)	0.9385(7)	0.6829(4)		0.3483(4)		
S-H	Orthorhombic	3.78a	0.5433(2)	0.9386(5)	0.6839(3)		0.3487(3)		
S-J	Hexagonal	None	0.54003(1)		0.71216(3)		0.17986(2)		
S-K	Hexagonal	None	0.53991(1)		0.71303(2)		0.18000(1)		
S-M	Hexagonal	2a and $3c$	0.54071(3)		0.7114(1)		0.1801(1)		
S-N	Hexagonal	2a and $3c$	0.54030(7)		0.7123(4)		0.1801(1)		
S-O	Hexagonal $2a$ and $3c$		0.53992(3)		0.7129(1)		0.1800(1)		

Figures in parentheses indicate standard deviations.

precession photographs of the crystals in S–J and S–K was consistent with hexagonal symmetry. The structure showed systematic absences $l \neq 2n$ for hh2hl and $l \neq 2n$ for 000l (Fig. 2). The probable space group is therefore $P6_3mc$, $P6_2c$ or $P6_3/mcc$. The cell dimensions (Table 3) and intensity distribution (Table 4) strongly suggest that these crystals are isostructural with α -C₂S.^{23),24)}

The crystals in S–M, S–N and S–O were hexagonal and showed two-dimensional modulations along the *a*-axis with N=2 and along the *c*-axis with N=3. The subcell dimensions determined from the main reflections of 102, 110, 201, 202, 212 and 300 are given in Table 3. The superstructure showed systematic absences $l \neq 2n$ for $h\bar{h}0l$, $l \neq 2n$ for $hh2\bar{h}l$ and $l \neq 2n$ for 000*l*; thus the probable space group is P6cc or P6/mcc. These crystallographic data are in accord with those given by Fukuda et al.,⁸⁾ who considered that their two-dimensionally modulated phase (x=0.398) was the transition product from the α -phase. The powder XRD patterns were very similar to that of the crystals composed of 66 mass% Ca₂SiO₄ and 34 mass% Ca₃(PO₄)₂ (x=0.364) by

Table 4. X-ray Powder Diffraction Data for Crystals S–J and S–K $(20^\circ{\le}2~\theta{\le}60^\circ,~Cu~K\alpha_1)$

h			S-J		S-K		
	к	ı	d _{calc} (nm)	Int.	d _{calc} (nm) Int.		
1	0	1	0.39092	3	0.39100 4		
0	0	2	0.35608	2	0.35651 2		
1	0	2	0.28331	27	0.28350 29		
1	1	0	0.27002	100	0.26995 100		
2	0	0	0.23384	2	0.23379 1		
2	0	1	0.22217	7	0.22215 8		
2	0	2	0.19546	13	0.19550 14		
0	0	4	0.17804	1	0.17826 2		
2	1	0	0.17677	<1	0.17673 <1		
2	0	3	0.16659	<1	0.16667 <1		
1	0	4	0.16639	<1	0.16656 <1		
2	1	2	0.15833	6	0.15834 6		
3	0	0	0.15589	6	0.15586 7		

The integrated intensities and unit-cell dimensions (Table 3) refined by the WPPD method. The reliability factors are $R_p=0.0781$, $R_{wp}=0.1004$ and $R_p(\text{peak})=0.1758$ for S-J, and $R_p=0.0777$, $R_{wp}=0.0990$ and $R_p(\text{peak})=0.1797$ for S-K.

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The subcell dimensions of the hexagonal phases (samples S–J to S–O) are plotted against x (Fig. 3). In the figure, the apparent subcell dimensions of the crystal in S–L ($a_s=0.54063(5)$ nm, $c_s=0.7120(3)$ nm and V=0.1802(1) nm³) are nearly intermediate between the dimensions of S–K (α -phase) and S–M (modulated hexagonal phase). Accordingly, the crystal must be the two-phase mixture. With the α - and modulated phases, the *a*-axis shrank and the *c*-axis expanded with increasing *x*. The volume increased for the modulated phase with increasing *x*.

4. Conclusions

(1) The phase constitution at ambient temperature, depending entirely on the P/(Si+P) ratio (=x), was classified into three categories according to the fraction of the α -to- $\alpha'_{\rm H}$ transition.

(2) Both the start and finish temperatures of the $\alpha'_{\rm L}$ -to- β martensitic transformation decreased with increasing *x*; 20°C < $M_{\rm f}$ for $x \le 0.080$, $M_{\rm f} < 20^{\circ}$ C < $M_{\rm s}$ for x = 0.100 and $M_{\rm s} < 20^{\circ}$ C for $x \ge 0.125$.

(3) The incommensurate phase showed a good correlation between modulation wavelength (=N) and x as N=4.134-1.56x $(0.175 \le x \le 0.250)$.

(4) With $0.275 \le x \le 0.300$, the quenched crystals were probably isostructural with the α -C₂S.

(5) The hexagonal phase with $0.350 \le x \le 0.400$ showed two-dimensional modulations (N=2 along the *a*-axis and N=3 along the *c*-axis), indicative of the transition product of the α -phase.

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References

- R. E. Barrett and W. J. McCaughley, Am. Mineral., 27, 680– 95 (1942).
- R. W. Nurse, J. H. Welch and W. Gutt, J. Chem. Soc., 220, 1077-83 (1959).
- W. Fix, H. Heymann and R. Heinke, J. Am. Ceram. Soc., 52, 346–47 (1969).
- 4) H. Saalfeld, Z. Kristallogr., 133, 396-404 (1971).
- 5) M. W. Barnes, M. Klimkiewicz and P. W. Brown, J. Am. Ceram. Soc., 75, 1423-29 (1992).
- H. Saalfeld and K. H. Klaska, Z. Kristallogr., 155, 65–73 (1981).
- K. Fukuda and I. Maki, J. Am. Ceram. Soc., 72, 2204–07 (1989).
- K. Fukuda, I. Maki, S. Ito, H. Yoshida and K. Aoki, J. Am. Ceram. Soc., 77, 2615–19 (1994).
- G. Lai, T. Nojiri and K. Nakano, Cem. Concr. Res., 22, 743– 54 (1992).



Fig. 3. Variation in subcell dimensions with P/(Si+P) ratio. a (A), c (B) and volume (C). (\Box) α -phase (Samples S–J and S–K). (\bigcirc) Hexagonal phase with two-dimensional modulations (S–M, S–N and S–O). (\heartsuit) The two-phase mixture (S–L). Error bars indicate 1σ .

- 10) K. Fukuda, I. Maki, K. Toyoda and S. Ito, J. Am. Ceram. Soc., 76, 1821-24 (1993)
- 11) K. Fukuda, I. Maki and S. Ito, J. Am. Ceram. Soc., 79, 2925-28 (1996).
- 12) K. Fukuda, I. Maki and S. Ito, J. Am. Ceram. Soc., 79, 2969-70 (1996).
- Y. J. Kim, I. Nettleship and W. M. Kriven, J. Am. Ceram. Soc., 75, 2407-19 (1992). 13)
- 14)K. Fukuda, I. Maki and K. Adachi, J. Am. Ceram. Soc., 75, 884-88 (1992).
- 15)
- H. Toraya, J. Appl. Crystallogr., 19, 440–47 (1986). F. Izumi, "The Rietveld Method," Oxford University Press 16)(1993) pp. 236-53.
- K. Fukuda and I. Maki, Cem. Concr. Res., 19, 913-18 (1989). 17)
- K. Fukuda and I. Maki, Cem. Concr. Res., 23, 599-602 18)

(1993).

- 19) J. W. Christian, "The Theory of Transformations in Metals and Alloys," Pergamon Press (1975).
- 20) G. Yamaguchi, Y. Ono, S. Kawamura and Y Soda, Yogyo-Kyokai-Shi, 71, 21-26 (1963).
- 21) D. L. Porter and A. H. Heuer, J. Am. Ceram. Soc., 62, 298-305 (1979).
- A. H. Heuer, N. Claussen, W. M. Kriven and M. Rühle, J. Am. Ceram. Soc., 65, 642-50 (1982). 22)
- 23) G. Yamaguchi, Y. Ono, S. Kawamura and Y. Soda, Yogyo-Kyokai-Shi, 71, 105-08 (1963).
- M. Regourd, M. Bigaré, J. Forest and A. Guinier, Proc. 5th 24)Int. Symp. Chem. Cement, Tokyo, Cement Association of Japan (1968) pp. 44-48.