Efficient synthesis of chrysanthemate precursor from chiral p-tolyl β -(trimethylsilyl)ethyl sulfoxide

PERKIN

Shuichi Nakamura, Yoshihiko Watanabe and Takeshi Toru*

Department of Applied Chemistry, Nagoya Institute of Technology, Gokiso, Showa-ku, Nagoya 466-8555, Japan. E-mail: toru@ach.nitech.ac.jp

Received (in Cambridge, UK) 24th September 1999, Accepted 1st November 1999

Reaction of α -lithio- β -silylethyl sulfoxide with ethyl 4-bromo- or 4-chloro-4-methylpent-2-enoate gives the cyclopropanecarboxylate as a single diastereomer which is converted to (1R)-trans-2-formyl-3,3-dimethylcyclopropanecarboxylate in high overall yield.

(1R)-trans-Chrysanthemate derivatives are more active than the (1S)-isomers to insects and less toxic to mammals. Asymmetric syntheses of chrysanthemate derivatives have been extensively studied;1 e.g. catalytic asymmetric cyclopropanations of 2,5dimethylhexa-2,4-diene with metal carbenoides using chiral ligands such as chiral Schiff bases, bisoxazolines, and salen complexes.2 In particular, Masamune et al. have succeeded in achieving a good level of trans- and enantioselectivity using chiral bis(4,5-diphenyl-1,3-oxazolinyl)methane.³ A diastereoselective synthesis of cis-chrysanthemates has been reported with moderate stereoselectivity through intramolecular cyclization using oxazolidinone as a chiral auxiliary.4 Recently, we reported a highly stereoselective β-addition reaction of the α-sulfinyl carbanion derived from β-(trimethylsilyl)ethyl sulfoxide to α,β-unsaturated carbonyl compounds, and stereoselective intramolecular cyclization. 5 We now report an efficient stereoselective synthesis of a chrysanthemate precursor, ethyl trans-formyl-3,3-dimethylcyclopropanecarboxylate, using ptolyl β -(trimethylsilyl)ethyl sulfoxide 1.

The reaction of α -lithio- β -silylethyl sulfoxide⁶ with ethyl 4-halo-4-methylpent-2-enoates was examined under various conditions. The results are summarized in Table 1.

A THF solution of p-tolyl β -(trimethylsilyl)ethyl sulfoxide 1 was treated with 1.25 equiv. of lithium diisopropylamide at −78 °C. After 5 min 1.3 equiv. of ethyl 4-bromo-4-methylpent-2-enoate was added rapidly and the mixture was stirred for 15 min. The reaction gave the cyclopropanecarboxylate 2† and the brominated sulfoxide 3[±] in 79 and 18% yields, respectively. Various reaction conditions have been tried to suppress the formation of 3 as shown in Table 1. When HMPA or copper iodide was added to the reaction mixture, the cyclization product 2 was obtained in 5 or 50% yield, respectively (entries 2 and 3). The reaction was carried out at −105 °C or at 0 °C giving 3 as a major product (entries 4 and 5). On the other hand, the cyclopropanecarboxylate 2 was formed in 84% yield without formation of the chlorinated compound when ethyl 4-chloro-4-methylpent-2-enoate was used (entry 6). The obtained cyclopropanecarboxylate 2 was found to be a single diastereomer. None of the other possible diastereomers was detected by ¹H NMR spectroscopy or HPLC analysis of the crude mixture. The trans configuration of 2 was assigned on the basis of the value of the vicinal coupling constant (5.4 Hz). Unfortunately, single crystals of 2 for the X-ray diffraction were not obtainable. Instead, the stereochemistry of the cyclopropanecarboxylate 4,5 obtained as a single diastereomer in a similar reaction between α-lithio-β-silylethyl sulfoxide and ethyl 4-bromobut-2-enoate, has been established by a single-crystal X-ray structure determination. Thus, the stereochemistry of 2 was reasonably assumed as $(R_s, 1'R, 1R, 2R)$ on the basis of the unequivocally confirmed stereochemistry of 4.

Table 1 Reaction of α-lithio- β -(trimethylsilyl)ethyl sulfoxide with ethyl 4-halo-4-methylpent-2-enoates

Entry	X	<i>T/</i> °C	Additive	Yield (%)	
				2	3
1	Br	-78	_	79	18
2	Br	-78	HMPA	5	60 a
3	Br	-78	CuI	50	45
4	Br	-105	_	24	64
5	Br	0	_	30	60
6	C1	-78	_	84	0

^a Sulfoxide 1 was recovered in 30% yield.

Remarkably, the chiral sulfoxide can completely control the stereochemistry of three consecutive chiral centers to be formed. The chiral sulfinyl group together with the β -trimethylsilyl group functions as a vinyl anion equivalent, not only inducing the stereoselectivity in the conjugate addition and cyclization reaction but also forming a double bond regioselectively from the product. Treatment of 2 with tetrabutylammonium fluoride in THF at room temperature afforded the β -eliminated compound 5 in 91% yield (Scheme 1). Thermal treatment of a benzene solution of 2 for 1 h in the presence of pyridine gave the product 6 as a mixture of E and E isomers in a ratio of 63:37 in 97% yield.

We confirmed that no epimerization occurred during the conversion of 2 into homoallylic carboxylates 5 and 6 by the vicinal coupling constants in the ¹H NMR spectra. Ozonolysis of the vinylsilane 6 was not successful under various conditions. The olefin 5 was treated with O₃ in a mixture of CH₂Cl₂ and MeOH at -78 °C for 3 h and subsequently with Me₂S to give ethyl (1*R*)-trans-2-formyl-3,3-dimethylcyclopropanecarboxylate 7 quantitatively. The chemical shifts and the coupling constants of the ¹H NMR spectrum and the IR spectral data of 7 were in good accord with those reported.⁷ The present procedure provides an efficient total synthesis of 1*R*-transchrysanthemates, since the aldehyde 7 can be easily transformed into them.⁸

J. Chem. Soc., Perkin Trans. 1, 1999, 3403-3404

Scheme 1

(1R)- trans- Chrysanthemate

Notes and references

 \dagger The compounds 2, 4, 5 and 6 gave satisfactory microanalytical and spectroscopic (IR, 1H NMR, ^{13}C NMR, m/z) data.

‡ Obtained in a diastereomer ratio of 59:41.

§ Selected data for 7: TLC $R_{\rm f}=0.69$ (Hexane–AcOEt = 70:30); $[a]_{\rm D}^{\rm 21.0}$ – 29.3 (c 0.246, acetone); $\delta_{\rm H}$ 1.25 (t, 3H, J = 7.1 Hz, -OCH₃), 1.32 (s, 3H, -CH₃), 1.40 (s, 3H, -CH₃), 2.41 (d, 1H, J = 4.5 Hz, -CO-CH-), 2.44 (dd, 1H, J = 2.0, 4.5 Hz, -CH-), 4.20 (q, 2H, J = 7.1 Hz, -OCH₂-), 9.60 (d, 1H, J = 2.0 Hz, -CHO); $\delta_{\rm C}$ 14.3, 20.5, 21.6, 26.0, 30.9, 33.4, 60.4, 171.6 IR (neat) 2970, 1730, 1460, 1450, 1400, 1250, 1200, 1130, 1100, 1030 cm⁻¹; SIMS m/z (rel. intensity) 170.1 (M⁺, 43), 149.1 (100), 113.0 (21), 57.0 (40), 28.0 (55); Anal. Calcd for $\rm C_9H_{14}O_3$: C, 63.51; H, 8.29; Found: C, 63.29; H, 8.51%.

- For reviews, see (a) D. Arlt, M. Jautelat and R. Lantzsch, Angew. Chem., Int. Ed. Engl., 1981, 20, 703; (b) J. Salaün, Chem. Rev., 1989, 89, 1247.
- 2 For recent reviews, see (a) V. K. Singh, A. DattaGupta and G. Sekar, *Synthesis*, 1997, 137; (b) M. P. Doyle and M. N. Protopopova, *Tetrahedron*, 1998, **54**, 7919.
- 3 (a) R. E. Lowenthal, A. Abiko and S. Masamune, *Tetrahedron Lett.*, 1990, **31**, 6005; (b) R. E. Lowenthal and S. Masamune, *Tetrahedron Lett.*, 1991, **32**, 7373.
- 4 W. A. Kleschick, M. W. Reed and J. Bordner, *J. Org. Chem.*, 1987, **52**, 3168.
- 5 T. Toru, S. Nakamura, H. Takemoto and Y. Ueno, *Synlett*, 1997, **5**, 449.
- 6 S. Kusuda, Y. Ueno and T. Toru, *Tetrahedron*, 1994, **50**, 1045.
- 7 (a) J. H. Babler and A. J. Tortorello, J. Org. Chem., 1976, 41, 885; (b)
 P. R. O. Montellano and S. E. Dinizo, J. Org. Chem., 1978, 43, 4323.
- 8 (a) M. J. Devos, L. Hevesi, P. Bayet and A. Krief, *Tetrahedron Lett.*, 1976, 3911; (b) W. G. Taylor, *Synthesis*, 1980, 554; (c) M. Fujita, T. Hiyama and K. Kondo, *Tetrahedron Lett.*, 1986, **27**, 2139; (d) H. Tanaka, S. Yamashita, M. Yamanoue and S. Torii, *J. Org. Chem.*, 1989, **54**, 444.

Communication 9/08649H