Correlation between a dielectric anomaly and a phase transition of sintered phosphorus doped WO₃ ceramics

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Phosphorus doped WO₃ ($P_xW_{1-x}O_3$) ceramics were fabricated by a solid state reaction. Temperature and frequency dependence of dielectric properties for the $P_xW_{1-x}O_3$ (x = 0-0.15) ceramics were evaluated at the temperature between -20 to 80° C. The dielectric constant of the $P_{0.05}W_{0.95}O_3$ ceramic was 3600-5800 at the frequency range from 1 kHz to 1 MHz in the room temperature, and the dielectric loss of the $P_{0.05}W_{0.95}O_3$ ceramic were less than 0.1 at the frequency of higher than 160 kHz. The dielectric anomaly peak of the WO₃ ceramic has been observed at 20°C, and the dielectric anomaly peak of the $P_{0.05}W_{0.95}O_3$ ceramic has been observed at about 0°C. The interaction between temperature dependence of dielectric properties (dielectric anomaly) ceramics and structural change of the WO₃ and $P_{0.05}W_{0.95}O_3$ were evaluated by Raman spectra. @2014 The Ceramic Society of Japan. All rights reserved.

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1. Introduction

Monoclinic tungsten trioxide (WO₃) shows a high dielectric constant, ¹⁾⁻⁴⁾ making WO₃ ceramic a candidate for use in ceramic capacitors. However, WO₃ resistivity is not high because oxygen defects exist in WO₃, producing electron carriers in WO₃^{5),6)} (see Eq. (1), with Kröger Vink notation).

$$O_0^x \to V_0^{\bullet \bullet} + 2e' + \frac{1}{2}O_2(gas)$$
(1)

In previous report, we have fabricated phosphorus doped WO₃ ceramics, and have evaluated the dielectric property of the $P_xW_{1-x}O_3$ ceramics.⁷⁾ When phosphorus with the valence of "+5" was substituted to W (the valence of "+6") site in WO₃, it assumed that the electron carriers in WO₃ were canceled out as follows Eq. (2).

$$P(\text{solid}) \xrightarrow{WO_3} P_W' + h^{\bullet}$$
(2)

Though the dielectric property of phosphorus doped WO₃ ($P_xW_{1-x}O_3$) ceramic could be evaluated because of its high electrical resistivity, the temperature dependence of $P_xW_{1-x}O_3$ ceramic has not been measured. In this investigation, we observed the temperature dependence of the dielectric property of non-doped WO₃ ceramics and $P_xW_{1-x}O_3$ ceramics. Both of the specimens were observed dielectric anomaly, and thus, we evaluated the correlation between the temperature dependence of a dielectric property and the structure of the non-doped WO₃ ceramics.

2. Experimental procedure

The starting materials were WO3 (Wako Pure Chemical

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Industries, Ltd., Japan) and $(NH_4)_2HPO_4$ (Wako Pure Chemical Industries, Ltd., Japan) powders. These powders were mixed with the stoichiometry *x* in $P_xW_{1-x}O_3$ between 0 to 0.15. The mixture was die-pressed to produce a pellet with 15 MPa. The pellet was calcined at 1000°C for 100 h in air. It was then ground and the PVA as binder was mixed; the powder was die-pressed again using uniaxial pressing with 15 MPa. The pellet was sintered at 1000°C for 24 h. Thereby; a phosphorus solid-soluted WO₃ specimen was synthesized.

The crystal structure of the specimens was determined using X-ray diffraction (XRD, Miniflex; Rigaku Corp.) at room temperature with Cu K α . Dielectric measurements were performed using an LCR meter (HP 4284A; Hewlett-Packard Inc.). Raman spectroscopy was carried out using a JASCO NR-1100 with a 532 nm line from an Nd:YAG laser.

Results and discussion

3.1 Dielectric property and temperature dependence

The $P_x W_{1-x} O_3$ (x = 0-0.15) powders were reacted at 1000°C for 100 h. **Figure 1** illustrates the XRD patterns for resulting specimens. All the samples with phosphorus contents x in $P_x W_{1-x} O_3$ between 0 and 0.15 were assigned as a monoclinic WO₃ single phase, and the other phases were not observed for all the specimens. These results were agreed well with the previous reports,⁷ and the resulting specimens could be used sufficiently for the electrical measurement.

Using these powders, the sintered $P_xW_{1-x}O_3$ (x = 0-0.15) pellets for electrical measurements were fabricated at 1000°C for 24 h. **Figure 2** presents dielectric properties of the WO₃ ceramics with phosphorus concentrations x of 0, 0.05, 0.10, and 0.15, and the measurement was carried out at room temperature of 20°C. The non-doped WO₃ indicated a huge-dielectric constant at the frequency between 1 kHz to 1 MHz, and the details were describes after. The dielectric loss of the non-doped WO₃ speci-

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Fig. 1. XRD patterns for the x in $P_xW_{1-x}O_3$ of 0–0.15 precursor powders.

men was very large between 1 kHz to 1 MHz, and the values were larger than 0.5. The dielectric loss of the $P_{0.05}W_{0.95}O_3$ and $P_{0.10}W_{0.90}O_3$ was less than 0.1 at frequencies higher than 100 kHz. The dielectric constants of the $P_{0.05}W_{0.95}O_3$ was 3600–5800 at 100–1000 kHz. Increasing the phosphorus contents in the WO₃ caused decreasing the dielectric constant and increasing the dielectric loss of the specimens. To measure the dielectric property, a small dielectric loss value causes precisely measuring a dielectric constant. Thereby, we focused and investigated the $P_{0.05}W_{0.95}O_3$ ceramic to evaluate a temperature dependence of the dielectric property.

Temperature dependence of dielectric properties for the WO₃ and the P_{0.05}W_{0.95}O₃ ceramic was measured at the temperature range between -20 to 80°C, and the profiles were shown in **Figs. 3** and **4**. The non-doped WO₃ ceramic indicated a dielectric anomaly at the temperature of around 20°C, and the P_{0.05}W_{0.95}O₃ ceramic showed a dielectric anomaly at the temperature of around 0°C. The reason of the non-doped WO₃ having the huge dielectric constant at room temperature (see Fig. 1) was because the measurement temperature of the non-doped WO₃. The dielectric anomaly of the non-doped WO₃ ceramic was attributed to the phase transition from the monoclinic δ phase to the triclinic



Fig. 2. Dielectric constant and loss of $P_x W_{1-x} O_3$ (0–0.15) ceramics at the frequency between 1 kHz to 1 MHz.



Fig. 3. Temperature dependence of the dielectric constant and the loss of the WO₃ ceramic.



Fig. 4. Temperature dependence of the dielectric constant and the loss of the $P_{0.05}W_{0.95}O_3$ ceramic.





Fig. 5. Raman spectra of WO_3 powder in the temperature range from -30 to 80° C with 10° C intervals.

Fig. 7. Raman spectra of $P_{0.05}W_{0.95}O_3$ powder in the temperature range from -30 to 80° C with 10° C intervals.



Fig. 6. Temperature dependence of (a) the peak positions and the (b) integrated intensities of Raman peaks B and E.

 γ phase.⁸⁾ Hirose et. al reported that WO₃ ceramic showed the dielectric anomaly at 17°C because of the phase transition, and our result was agreed well with the report.⁹⁾ It assumed that the dielectric anomaly of the P_{0.05}W_{0.95}O₃ ceramic was also attributed to the phase transition of WO₃ (phosphorus doped WO₃), and we investigated the correlation of the dielectric anomaly and the structure by Raman spectroscopy on the next section. For both of samples, increasing temperature caused increase of the dielectric anomaly temperature.

3.2 Raman spectra

In order to confirm between the dielectric property and the

phase transition, Raman spectroscopy measurement was carried out for the WO_3 and the $P_{0.05}W_{0.95}O_3$ powder between -30 to $80^\circ C.$

Figure 5 portrays the Raman spectra for the WO₃ powder. The peaks were observed at 130 (A), 264 (B), 320 (C), 701 (D) and 791 (E) cm⁻¹ for all the temperatures. These peaks were attributed to the WO₃ monoclinic phase (γ) or the triclinic phase (δ).¹⁰ The peak A was associated to O–O deformation,¹¹ the peaks B and C were assigned to W–O deformations,¹² the peaks D and E were attributed to W–O stretching modes.¹² All the peaks were sharpened and shifted to the position at the higher wavenumber position between 20–30°C, in particular, the peaks B and E changed remarkably. **Figure 6** represents the positions and



Fig. 8. Temperature dependence of (a) the peak positions and the (b) integrated intensities of Raman peaks B' and E'.

the integrated intensity of the peaks B and E as a function of temperature. Beyond the temperature of $20-30^{\circ}$ C, both of the peak positions shifted to high wavenumber, and the integrated intensities enhanced drastically. These change assumed to be the crystallographic transition of the WO₃, and the phase transition temperature agreed well with the dielectric anomaly temperature.

Figure 7 displays the Raman spectra for the $P_{0.05}W_{0.95}O_3$ powder. The peaks were observed at 132 (A'), 265 (B'), 323 (C'), 705 (D') and 794 (E') cm^{-1} as well as the result of the nondoped WO₃, and these peaks were speculated to be the γ phase or the δ phase.¹⁰⁾ All the peaks were sharpened and shifted to the position at the lower wavenumber position between $-20-0^{\circ}$ C, in particular, a remarkable change was observed for the peaks B' and E'. Figure 8 represents the positions and the integrated intensity of the peaks B' and E' as a function of temperature. Both of the peak positions shifted to low wavenumber at the temperature of -20--10°C, and the integrated intensities increased drastically at the temperature of -10-0°C. These results suggested that the phase transition of $P_{0.05}W_{0.95}O_3$ occurred at the temperature between $-20-0^{\circ}$ C. It assumed that the structural phase transition started at -20--10°C and occurred remarkably at $-10-0^{\circ}$ C. The results were agreed with the dielectric anomaly of the P_{0.05}W_{0.95}O₃ ceramic. Furthermore, the results of Raman spectra and dielectric anomaly suggested that phosphorus doping in WO₃ host causes decrease of the dielectric anomaly temperature.

4. Conclusion

Dielectric property and Raman spectroscopy observation was carried out for the $P_xW_{1-x}O_3$ (x = 0-0.15) ceramics at various

temperatures. The WO₃ ceramic and the $P_{0.05}W_{0.95}O_3$ ceramic indicated a dielectric anomaly at 17 and 0°C, respectively. By Raman spectra observation, the structural phase transition of the WO₃ ceramic and the $P_{0.05}W_{0.95}O_3$ ceramic were at the temperature of 20–30°C and -20–0°C, respectively. Thus, WO₃ dielectric anomaly temperature associating to phase transition decreased with increasing phosphorus contents in $P_xW_{1-x}O_3$.

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