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PERSPECTIVES





Hard-sphere melting and crystallization with event-chain Monte Carlo

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We simulate crystallization and melting with local Monte Carlo (LMC), with event-chain Monte Carlo (ECMC), and with event-driven molecular dynamics (EDMD) in systems with up to one million three-dimensional hard spheres. We illustrate that our implementations of the three algorithms rigorously coincide in their equilibrium properties. We then study nucleation in the NVE ensemble from the fcc crystal into the homogeneous liquid phase and from the liquid into the homogeneous crystal. ECMC and EDMD both approach equilibrium orders of magnitude faster than LMC. ECMC is also notably faster than EDMD, especially for the equilibration into a crystal from a disordered initial condition at high density. ECMC can be trivially implemented for hard-sphere and for soft-sphere potentials, and we suggest possible applications of this algorithm for study-ing jamming and the physics of glasses, as well as disordered systems. © 2015 AIP Publishing LLC. [http://dx.doi.org/10.1063/1.4929529]

I. INTRODUCTION

Crystallization and melting have long been central subjects in statistical physics. These processes connect microscopic nucleation with the macroscopic phenomena of domain growth and of phase transitions. A number of numerical methods and simulation techniques have been brought to bear on these subjects, following the pioneering computer simulations of hard-sphere systems by both Monte Carlo^{1–3} and by molecular dynamics.^{4–8}

The hard-sphere system is trivial to describe. Nevertheless, equilibration in this simplest of all particle systems is a slow process, because of the large activation free energy for crystallization. Timescales are also especially large in the fluid-solid coexistence regime, because of the surface tension between coexisting phases. Specialized algorithms for equilibration have been developed to overcome these problems, and the melting and crystallization time scales provide useful benchmarks for their comparison.

A rejection-free hard-sphere "event-chain" Monte Carlo algorithm (ECMC)⁹ has recently allowed to speed up equilibration for two-dimensional hard disks by roughly two orders of magnitude compared to the event-driven molecular dynamics^{5,10} (EDMD) and to local Monte Carlo^{11,12} (LMC). In ECMC, a randomly sampled starting sphere moves along a straight line until the latter collides with another sphere, which then moves in the same direction until it collides itself with yet another sphere. This continues until the spheres' total displacement equals a certain fixed length L_c . ECMC breaks detailed balance (moves are in the +x, +y, and +z directions only) yet satisfies global balance and ergodicity.¹³ It rigorously samples the equilibrium Boltzmann distribution. Considerable speedup was also demonstrated for the extension of ECMC to continuous potentials.^{13,14}

In this paper, we assess the speed of ECMC, EDMD, and LMC not by computing autocorrelation functions in equilibrium, but rather by the time scales associated with melting and crystallization in systems of many spheres at high density. We expect our observations to extend to subjects as dense packing, nucleation, and jamming. We focus on the melting from the metastable solid branch to the stable liquid (that is, slightly below the liquid-solid coexistence interval) and also the nucleation processes from the metastable liquid branch towards the stable solid slightly above coexistence. The paper is organized as follows: Our model system and our methods are described in Section II, together with the observables on which we focus: The non-dimensional pressure and the local and global orientational order parameters. Results are summarized in Section III: We reproduce the phase diagram by ECMC and quantify the efficiencies of our three methods. We discuss the relative efficiency of ECMC and EDMD for the crystallization process. Concluding remarks are described in Section IV.

II. MODEL, ALGORITHMS, AND OBSERVABLES

We consider *N* monodisperse hard spheres of radius σ in a cubic box of sides *L* with periodic boundary conditions (PBC). The density (packing fraction) ν is given by $\nu = 4/3N\pi\sigma^3/L^3$. We concentrate on melting and crystallization from the unstable to the stable phase, i.e., from the unstable crystalline branch to the liquid and from the unstable liquid branch to the crystal. For our melting runs, at density $\nu = 0.490$ below the coexistence interval of liquid and solid phases, we prepare the initial configurations as perfect fcc crystals, corresponding to the stable phase at high densities (the free energy difference between fcc and hcp crystals has been discussed actively in

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Refs. 15 and 16). The fcc initial conditions are compatible with the cubic simulation box.

For the crystallization runs at density v = 0.548 above the coexistence interval, we start from a fluid initial configuration at a liquid-phase density v = 0.490. In order to reach the higher target density, we repeatedly increase σ slightly and remove all created overlaps by sliding overlapping pairs of spheres for a half-length of overlap along their common symmetry axis. This is done until all pair overlaps have disappeared.

In EDMD, hard spheres evolve in continuous physical time through collisions, and the dynamics solves Newton's classical equations of motion. We use an efficient sequential implementation.¹⁰ LMC and ECMC are implemented very simply. For the former, the optimal displacement of spheres is determined by short LMC runs from the initial conditions so that the acceptance ratio is 1/2. For the latter, a single parameter, the chain length L_c , must be optimized for each density. A single event can be implemented very quickly in ECMC, as the motion is always in +x or +y, which decreases the CPU time per event. We consider systems with N = 2048, 131072,and 1048576 spheres. Our calculations are mainly done on the Intel Xeon CPU E5-2680 2.80 GHz, where we reach ~3.15 $\times 10^9$ events/h of CPU time for ECMC and $\sim 4.62 \times 10^8$ collisions/h for EDMD. For the LMC algorithm, $\sim 6.5 \times 10^9$ trials/h are reached. All our comparisons of algorithms are in terms of CPU time. To be as fair as possible, the three algorithms were implemented following unified design principles. Furthermore, we used the same computer, the same language (Intel FORTRAN), and the same optimal option of compiler. An event-count would have produced similar results.

We track the time-evolution of the ordering/disordering of the system from the pressure and the local and global orientational order parameters. In EDMD, the nondimensional virial pressure is computed from the collision rate via the virial theorem,

$$P^* = \beta P (2\sigma)^3 = \frac{6\nu}{\pi} \left[1 - \frac{\beta m}{3T} \frac{1}{N} \sum_{\text{collisions}} b_{ij} \right], \quad (1)$$

where T is the total simulation time, and $\beta = 1/m \langle v_x^2 \rangle$ is the inverse kinetic temperature (mass m and mean-square *x*-component of velocity of spheres). The collision force $b_{ij} = \mathbf{r}_{ij} \cdot \mathbf{v}_{ij}$ is defined between the relative positions and the relative velocities of the collision partners.^{11,17} In ECMC, the pressure P^* can be evaluated from the mean excess chain displacement,¹³

$$P^* = \frac{6\nu}{\pi} \left\langle \frac{x_{\text{final}} - x_{\text{initial}}}{L_c} \right\rangle_{\text{chains}},\tag{2}$$

where x_{final} and x_{initial} are final and initial positions of each chain, respectively, taking into account the PBC. This convenient formula replaces the tedious extrapolation of the pair correlation function at contact $g_2(r = 2\sigma)$ that was used previously and that must still be used for LMC. The comparison of the evolving pressure of EDMD (using Eq. (1)) and ECMC (using Eq. (2)) in the liquid state at v = 0.490 is shown in the left of Fig. 1. The pressures fluctuate around $P^* = 11.3893$ and agree within very tight error bars.

Besides the pressure, we quantify the speed of melting and of crystallization via the time-dependent local q_6 and global Q_6 order parameters,¹⁸

$$q_{6} = \frac{1}{N} \sum_{i=1}^{N} \sqrt{\frac{4\pi}{13} \sum_{m=-6}^{m=6} \left| \frac{1}{n(i)} \sum_{j=1}^{n(i)} Y_{6,m}(\mathbf{r}_{ij}) \right|^{2}}, \qquad (3)$$

$$Q_6 = \sqrt{\frac{4\pi}{13}} \sum_{m=-6}^{m=6} \left| \frac{1}{Nn(i)} \sum_{i=1}^{N} \sum_{j=1}^{n(i)} Y_{6,m}(\mathbf{r}_{ij}) \right|^2,$$
(4)

where n(i) is the number of nearest neighbors for each sphere *i* and $Y_{6,m}(\mathbf{r}_{ij})$ are the spherical harmonics with icosahedral symmetry for the distance vector \mathbf{r}_{ij} between spheres *i* and *j*. We detect nearest neighbors by the SANN algorithm of van Meel *et al.*¹⁹ rather than by the Voronoi construction. On the right of Fig. 1, we again demonstrate that the equilibrium values for $q_6(\sim 0.38678)$ agree within tight error bars for LMC, ECMC, and EDMD. Similar agreement was reached for the global order parameter Q_6 . We note that perfect fcc configurations have an orientational order $Q_6 = q_6 \sim 0.575$, while for liquid configurations (including our disordered initial configurations), Q_6 approaches zero. In the dense liquid, the



FIG. 1. Pressure P^* (left) and local order parameter q_6 (right) in the dense liquid at v = 0.490, obtained from ECMC and EDMD for $N = 131\,072$ hard spheres. In ECMC, the pressure is computed using the excess displacement method of Eq. (2). In the right panel, the local order parameter of LMC is also shown.

local order q_6 is non-zero because of the buildup of transient local crystal structures.²⁰

III. RESULTS

A. Hard-sphere phase diagram

The fluid-solid coexistence in the NVE ensemble (which, for hard spheres, corresponds to the common NVT ensemble) for densities v in the interval 0.494 < v < 0.545 has been investigated for more than 50 years.^{2,4,7} Recently, various theoretical equations of states were compared with the results of a large-scale EDMD simulation on this system,²¹ with N $\sim 10^6$. The metastable fluid branch in the fluid-solid coexistence window was found stable against freezing on EDMD time scales up to $\sim O(10^9)$ collisions. To speed up the simulation, the replica exchange MC method was adapted to the hardsphere case. To keep the acceptance rate at reasonable values, many replicas at finely spaced densities had to be used, and this approach proved restricted to quite small system sizes (N = 32and N = 108).²² Fernández *et al.*²³ explored the coexistence of hard-sphere systems in equilibrium by tethered MC for relatively small system sizes $\sim O(10^3)$. In this method, the approach to equilibrium is accelerated by a biased field of two order parameters. The equilibrium pressure $P^* = 11.5727(10)$ is obtained through extrapolation towards the infinite-size limit. We note that in three-dimensional hard spheres, the direct simulation remains difficult in the coexistence region, even for ECMC, whereas for the analogous two-dimensional hard disks, the equilibration of the coexisting hexatic and liquid phases by ECMC proved possible at all densities, for up to one million disks.²⁴



FIG. 2. Hard-sphere equation of state obtained by ECMC after $\sim O(10^{12})$ collisions. The snapshots in the coexistence interval at N = 1.048576 represent the local orientational order $q_6(i)$ for each sphere *i* (liquid-like local order is represented in green-solid-like local order in blue). The equilibrium coexistence pressure for the infinite system is also shown. The densities $\nu = 0.490$ and $\nu = 0.548$, on which we concentrate in Sections III B and III C, are indicated.

Fig. 2 shows the hard-sphere equation of state from ECMC (final pressures after long runs with $(2 \sim 3 \times 10^{12})$ collisions). The pressure is averaged over 10^{10} collisions at the end of the simulation. The stable and unstable fluid pressures well agree with Hoover-Ree²⁵ and Carnahan-Starling extrapolation.²⁶ Inside the coexistence interval, the final pressure depends on the initial configuration, as the metastable fcc solid or fluid initial configurations are preserved on the time-scale of the simulation. In the NVE ensemble, the presence of interfaces of different topologies makes that the equation of state is non-monotonous, and the liquid-solid coexistence pressure curve is not flat in a finite system. As one increases the density from the liquid phase, the spherical or cylindrical droplets that can be seen in Fig. 2 generate an excess Laplace pressure. This is analogous to what was found in two-dimensional hard disks,^{24,27} which show droplets and stripes or in fluid-gas mixtures of the three-dimensional Lennard-Jones system,²⁸ where spherical and cylindrical droplets as well as two-dimensional stripes are found.

Specifically, for v < 0.498, simulations from arbitrary initial conditions converge to the same pressure since the system is completely liquid and nucleation barriers are low. In the region of $v = 0.500 \sim 0.514$, simulations from fcc initial configurations successfully create interfaces with different topologies, whereas simulations from fluid initial conditions remain on the fluid branch. The pressure at $v = 0.498 \sim 0.514$ decreases as $P^* = 12.2 \sim 11.5$ and agrees with the expected coexistence pressure.²³ The phase coexistence at equilibrium can be seen clearly by the spatial distributions of the local q_6 order parameter, where the fcc crystal reduces to a droplet $(v = 0.500 \sim 0.502)$ when started from a fcc crystal. For larger densities ($\nu = 0.504 \sim 0.512$), the remaining fcc phase has the form of a cylinder that reconnects through the PBC, surrounded by the liquid dominant phase created through melting (see the insets of Fig. 2). In the density interval $v = 0.514 \sim 0.530$, the fcc crystal and the liquid remain metastable on the available scales of simulation time. For $v = 0.532 \sim 0.543$, simulations from fluid initial conditions nucleate fcc droplets. At v> 0.543, simulations from fluid initial condition drop down



FIG. 3. Melting at $\nu = 0.490$ (N = 131072) from a fcc initial configuration into the stable liquid, tracked by the time evolution of the global Q_6 order parameter in LMC, EDMD, and ECMC with optimal chain length ($L_c/2\sigma$ = 5.87). Data averaged over 5 samples. Note that ECMC and EDMD are orders of magnitude faster than LMC.



FIG. 4. Crystallization at v = 0.548 from a random initial configuration, tracked by the evolution of the pressure P^* in EDMD and ECMC with different chain lengths L_c . Data averaged over 100 samples for N = 131072 (left) and 5 samples for N = 1048576 (right). The inset in the left panel illustrates the influence of the parameter L_c on the performance of ECMC.

near the solid branch, however, relaxation is in progress at the value around slightly higher pressure than the solid branch, except for the case of N = 2048.

B. Melting from a fcc initial configuration at a fluid density

We now study the speed of melting into the equilibrium liquid phase at v = 0.490 from a fcc crystalline initial configuration for N = 131072. From the global order parameter Q_6 , as shown in Fig. 3, the initial fcc configuration $(Q_6 \sim 0.575)$ rapidly becomes unstable for the three algorithms and approaches the liquid branch, where the global orientational order approaches zero. Melting is much faster for ECMC and EDMD than for LMC. ECMC is found to be somewhat faster than EDMD.

C. Crystallization from a fluid initial condition at a solid density

In Fig. 4, we show the evolution toward crystallization of EDMD and of ECMC with different chain lengths L_c . Results

are averaged over 100 samples for N = 131072 (left of Fig. 4) and over 5 samples for N = 1048576 (right of Fig. 4). Results for three trial runs in which we changed the chain length L_c are also shown Fig. 4. The efficiency of ECMC naturally depends on L_c . For both methods, the pressure remains somewhat above the configurational equilibrium pressure $P^* \sim 11.934$. The relative advantage of ECMC with optimal chain length is evident.

In Figs. 5 and 6, we show the evolution of the local q_6 and global Q_6 order parameters in crystallization runs of EDMD and ECMC with three chain lengths. The number of samples is again 100 for N = 131072 and 5 for N = 1048576. In the early stage of relaxation, q_6 and Q_6 are increasing functions of CPU time. In the perfect fcc configuration, the local and global order parameters agree to $q_6 = Q_6 = 0.57452$. Due to thermal fluctuations, actual numerical simulation at fcc equilibrium estimates the order parameters to $(q_6, Q_6) = (0.505, 0.483)$ in $(N, \nu) = (1048576, 0.548)$. Although the crystallization still proceeds, our final averaged q_6 and Q_6 reach around 0.488 and 0.432, respectively. The inconsistency between the fcc crystal structure and the (finite) simulation box may well be responsible for the reduction in order parameter.



FIG. 5. Crystallization at v = 0.548 from a fluid initial configuration with N = 131072, tracked by the evolution of local q_6 and global Q_6 order parameters by LMC, EDMD, and ECMC with different chain lengths L_c . Data averaged over 100 samples.



FIG. 6. Crystallization at $\nu = 0.548$ from a random initial configuration with N = 1048576 tracked by the evolution of local q_6 and global Q_6 order parameters in LMC, EDMD, and ECMC with different chain lengths L_c . Data averaged over 5 samples. Note that LMC is much slower than the other two methods.

Order parameters decay towards higher order in time, and ECMC with optimal chain length $L_c/(2\sigma) = 23.46$ needs 48.6 CPU hours to reach at $q_6 = 0.4863$ in N = 131072, however EDMD needs 165 CPU hours. Both smaller and larger L_c result in the inefficiency of relaxation. After 165 h, one quarter of ECMC simulations had almost reached the equilibrium state (i.e, $q_6 > 0.99 \times q_{6ea}$) whereas for EDMD, only 10% had reached such values. The results for LMC are also shown in Fig. 5, and they show that it is much slower than ECMC and EDMD. For example, our LMC simulation on average reaches $q_6 = 48.25$ after $O(10^7)$ LMC sweeps corresponding to 152 CPU hours. At this $q_6 = 48.25$, ECMC and EDMD need only about 11 and 28 CPU hours, respectively. The decay of Q_6 has the same tendency as that of q_6 ; however, its values are lower than that of q_6 . Since the whole system is slower to order than to establish local order, the global orientation also grows more slowly than the local orientation. In the larger case N= 1048576, ECMC with optimized chain length and EDMD need 416 and 1000 CPU hours to reach $q_6 = 0.4897$, respectively. Note that if chain length is not optimized, the performance of ECMC changes drastically and becomes comparable

to or slower than that of EDMD. As for q_6 and Q_6 , ECMC with optimal length is faster than that of EDMD for a certain factor depending on the target point, which will be discussed in Section III D.

D. Relative speed of ECMC and EDMD

To further quantify the equilibration speed of ECMC and EDMD, rather than the observables as a function of time O(t), we consider the elapsed CPU time $\mathcal{T}(O)$ from the beginning of the simulation t = 0 at which the observable O is reached. This allows us to define the relative efficiency R_s ,

$$R_s(O) = \mathcal{T}_{\text{EDMD}}(O) / \mathcal{T}_{\text{ECMC}}(O), \tag{5}$$

and analogously for any pair of algorithms, where O is an observable, in our case the pressure P^* , or the local and global order parameters q_6 and Q_6 .

For the melting case (not shown) at v = 0.490, R_s for observables takes around 3.2 (t = 0) to 1 at the end of simulation. EDMD is slightly slower than that of ECMC. This relative speed remains rather unchanged during the melting process.



FIG. 7. Relative speed R_s (see Eq. (5)) of LMC, ECMC and EDMD as a function of normalized observables $\hat{O} = (O - O_{initial})/(O_{equil.} - O_{initial})$, (left) $O = P^*$ and (right) $O = Q_6$ for N = 1048576. As also shown in Fig. 6, both ECMC and EDMD are orders of magnitude faster than LMC. ECMC shows considerable advantage over EDMD in the later times of the evolution, when large-scale structures are built up.

The situation changes drastically for the crystallization process. Fig. 7 shows R_s as a function of normalized observables as $\hat{O} = (O - O_{\text{initial}})/(O_{\text{equil.}} - O_{\text{initial}})$ at $\nu = 0.548$ in the crystallization process, at which each data can be estimated by Figs. 4–6. O_{initial} and $O_{\text{equil.}}$ are the values at t = 0 and at the equilibrium, respectively. Those are obtained by independent runs at the perfect fcc crystal which are $q_6 = 0.505$, Q_6 = 0.483, and P^* = 11.934. In all cases, $R_s(O)$ is increasing function and growing drastically near the crystal (i.e., $\hat{O} > 0.8$) as a *hockey stick* curve. The relative efficiency R_s depends rather weakly on system size. We did not compute variations precisely, as the running times \mathcal{T} were only averaged over 5 samples for N = 1048576. The different algorithmic complexities of our methods might marginally contribute to the size dependence of R_s : Our EDMD algorithm is implemented in $O(N \log N)$ per N collisions and ECMC as O(N) per N collisions. At $q_6 = 0.47$, $R_s(q_6)$ takes 1.33 (N = 131072) and 1.49 (N = 1048576), respectively.

IV. CONCLUSION

In this paper, we compared hard-sphere Monte Carlo and molecular dynamics algorithms that coincide in their equilibrium properties. In large systems with up to one million spheres, we recovered the known phase diagram and especially the coexistence region. We quantified the approach towards equilibrium, namely, towards the fcc crystal from the liquidlike initial configuration at packing v = 0.548 or the stable liquid from a fcc initial configuration at packing v = 0.490. We clearly showed that the EDMD and ECMC are orders of magnitude faster than the LMC algorithm for both the melting and the crystallization. ECMC needs optimization for chain length L_c , and we generally find that the individual chains should wrap a few times around the simulation box. The effect of the chain length is rather drastic, and L_c must be optimized carefully. The optimal chain length for the crystallization process is estimated around $L_c/(2\sigma) \sim 25$ (N = 131072) and 50 (N = 1.048576). With a fixed $L_c/(2\sigma)$, the actual chain length $\langle x_{\text{final}} - x_{\text{initial}} \rangle / (2\sigma)$ can be obtained by trial and error before the production runs. It may also be estimated by the ECMC pressure formula, Eq. (2), as

$$\frac{\langle x_{\text{final}} - x_{\text{initial}} \rangle}{2\sigma} = \frac{P^*(L_c/(2\sigma))\pi}{6\nu} \tag{6}$$

which is evolving during simulation according to pressure relaxation. In case of the optimal chain length $L_c/2\sigma = 23.46 \sim L_x/2$, the chain winds around 6 times around the periodic box. While doing so, very few spheres get hit more than once.

We conclude that ECMC with well-chosen chain lengths is far superior to LMC, although it can be implemented just as easily.^{29,30} Even with respect to molecular dynamics, it performs very well. The clearest advantage of ECMC over EDMD shows up in the crystallization, that is, in the buildup of longrange correlations. We expect the ECMC algorithm and its extension to continuous potentials to be helpful to investigate jamming and to estimate accurate nucleation rate^{31,32} and to analyze the full scenario of nucleation and precursor crystallization.³³ Of particular interest might be that ECMC remains event-driven even for continuous potentials and very simple to implement. Molecular dynamics, on the other hand, must be implemented as a time-driven algorithm for continuous potentials. The discretization of the equations of motion then makes molecular dynamics rather awkward to implement.

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- ¹N. Metropolis, A. W. Rosenbluth, M. N. Rosenbluth, A. H. Teller, and E. Teller, J. Chem. Phys. **21**, 1087 (1953).
- ²W. W. Wood and J. D. Jacobson, J. Chem. Phys. 27, 1207 (1957).
- ³W. Krauth, *Statistical Mechanics: Algorithms and Computations* (Oxford University Press, Oxford, 2006).
- ⁴B. J. Alder and T. E. Wainwright, J. Chem. Phys. 27, 1208 (1957).
- ⁵B. J. Alder and T. E. Wainwright, J. Chem. Phys. **31**, 459 (1959).
- ⁶B. J. Alder and T. E. Wainwright, Phys. Rev. 127, 359 (1962).
- ⁷B. J. Alder, W. G. Hoover, and D. A. Young, J. Chem. Phys. 49, 3688 (1968).
- ⁸The 50th Anniversary of the Alder Transition: Recent Progress on Computational Statistical Physics — Proceedings of the Symposium on the 50th Anniversary of the Alder Transition, edited by Y. Hiwatari and M. Isobe [Prog. Theor. Phys. Suppl. **178**, 1-192 (2009)]; available at http://ptps. oxfordjournals.org/content/178.toc.
- ⁹E. P. Bernard, W. Krauth, and D. B. Wilson, Phys. Rev. E **80**, 056704 (2009). ¹⁰M. Isobe, Int. J. Mod. Phys. C **10**, 1281 (1999).
- ¹¹M. Engel, J. A. Anderson, S. C. Glotzer, M. Isobe, E. P. Bernard, and W.
- Krauth, Phys. Rev. E 87, 042134 (2013).
- ¹²E. P. Bernard and W. Krauth, Phys. Rev. E **86**, 017701 (2012).
- ¹³M. Michel, S. C. Kapfer, and W. Krauth, J. Chem. Phys. **140**, 054116 (2014).
- ¹⁴S. C. Kapfer and W. Krauth, Phys. Rev. Lett. **114**, 035702 (2015).
- ¹⁵L. V. Woodcock, Nature **385**, 141 (1997).
- ¹⁶P. B. Bolhuis, D. Frenkel, S. Mau, and D. Huse, Nature 388, 235 (1997).
- ¹⁷J. J. Erpenbeck and W. W. Wood, in *Statistical Mechanics Part B. Modern Theoretical Chemistry*, edited by B. J. Berne (Plenum, New York, 1977), Vol. 6, Chap. I, p. 1.
- ¹⁸P. J. Steinhardt, D. R. Nelson, and M. Ronchetti, Phys. Rev. B 28, 784 (1983).
- ¹⁹J. A. van Meel, L. Fillion, C. Valeriani, and D. Frenkel, J. Chem. Phys. **136**, 234107 (2012).
- ²⁰M. Isobe and B. J. Alder, J. Chem. Phys. **137**, 194501 (2012).
- ²¹M. N. Bannerman, L. Lue, and L. V. Woodcock, J. Chem. Phys. **132**, 084507 (2010).
- ²²G. Odriozola, J. Chem. Phys. 131, 144107 (2009).
- ²³L. A. Fernández, V. Martín-Mayor, B. Seoane, and P. Verrocchio, Phys. Rev. Lett. **108**, 165701 (2012).
- ²⁴E. P. Bernard and W. Krauth, Phys. Rev. Lett. **107**, 155704 (2011).
- ²⁵W. G. Hoover and F. H. Ree, J. Chem. Phys. **49**, 3609 (1968).
- ²⁶N. F. Carnahan and K. E. Starling, J. Chem. Phys. **51**, 635 (1969).
- ²⁷J. E. Mayer and W. W. Wood, J. Chem. Phys. 42, 4268 (1965).
- ²⁸K. Binder, B. J. Block, P. Virnau, and A. Troester, Am. J. Phys. **80**, 1099 (2012).
- ²⁹S. C. Kapfer and W. Krauth, J. Phys.: Conf. Ser. **454**, 012031 (2013).
- ³⁰T. A. Kampmann, H. H. Boltz, and J. Kierfeld, J. Comput. Phys. 281, 864 (2015).
- ³¹L. Filion, M. Hermes, R. Ni, and M. Dijkstra, J. Chem. Phys. **133**, 244155 (2010).
- ³²S. Auer and D. Frenkel, Nature 409, 1020 (2001).
- ³³L. Gránásy and G. I. Tóth, Nat. Phys. 10, 12 (2014).